

# Androsta-2,4-diene-3,17-diol, 17-acetate 3-(2,2,3,3,4,4,4-heptafluorobutanoate), (17«beta»)-

**Other names:** 17-(Acetyloxy)androsta-2,4-dien-3-yl 2,2,3,3,4,4,4-heptafluorobutanoate, (17«beta»)-; Testosterone, 3-HFB, 17«beta»-Ac.

**InChI:** InChI=1S/C25H29F7O4/c1-13(33)35-19-7-6-17-16-5-4-14-12-15(8-10-21(14,2)18(16)9-11-22(17,19)3)36-20(34)23(26,27)24(28,29)25(30,31)32/h8,12,16-19H,4-7,9-11H2,1-3H3/t16-,17-,18-,19-,21-,22-/m1/s1

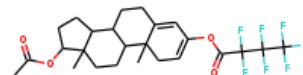
**InChI Key:** NTZWSZQHFGGWDX-OYSQMQRMSA-N

**Formula:** C<sub>25</sub>H<sub>29</sub>F<sub>7</sub>O<sub>4</sub>

**SMILES:** CC(=O)OC1CCC2C3CCC4=CC(OC(=O)C(F)(F)C(F)(F)C(F)(F)F)=CCC4(C)C3CCC12C

**Molecular Weight:** 526.48

**CAS:** 49566-69-0



## Physical Properties

Property	Value	Unit	Source
$\Delta_f G^\circ$	-1466.61	kJ/mol	Joback Method
$\Delta_f H^\circ_{\text{gas}}$	-2105.13	kJ/mol	Joback Method
$\Delta_{\text{fus}} H^\circ$	38.65	kJ/mol	Joback Method
$\Delta_{\text{vap}} H^\circ$	79.45	kJ/mol	Joback Method
$\log P_{\text{oct/wat}}$	6.751		Crippen Method
$P_c$	1058.95	kPa	Joback Method
$T_{\text{boil}}$	956.91	K	Joback Method
$T_c$	1176.82	K	Joback Method
$T_{\text{fus}}$	647.26	K	Joback Method
$V_c$	1.331	m <sup>3</sup> /kg-mol	Joback Method

## Temperature Dependent Properties

Property	Value	Unit	Temperature (K)	Source
$C_{p,\text{gas}}$	1236.02	J/mol×K	956.91	Joback Method

## Sources

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**NIST Webbook:** [http://webbook.nist.gov/cgi/inchi/InChI=1S/C25H29F7O4/c1-13\(33\)35-19-7-6-17-16-5-4-14-12-15\(8-10-21\(14,2\)18\(16\)9-11-22\(17,19\)3\)36-20\(34\)23\(26,27\)24\(28,29\)25\(30,31\)32/h8,12,16-19H,4-7,9-11H2,1-3H3/t16-,17-,18-,19-,21-,22-/m1/s1](http://webbook.nist.gov/cgi/inchi/InChI=1S/C25H29F7O4/c1-13(33)35-19-7-6-17-16-5-4-14-12-15(8-10-21(14,2)18(16)9-11-22(17,19)3)36-20(34)23(26,27)24(28,29)25(30,31)32/h8,12,16-19H,4-7,9-11H2,1-3H3/t16-,17-,18-,19-,21-,22-/m1/s1)

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

$C_{p, gas}$ : Ideal gas heat capacity (J/mol×K).

$\Delta_f G^\circ$ : Standard Gibbs free energy of formation (kJ/mol).

$\Delta_f H^\circ_{gas}$ : Enthalpy of formation at standard conditions (kJ/mol).

$\Delta_{fus} H^\circ$ : Enthalpy of fusion at standard conditions (kJ/mol).

$\Delta_{vap} H^\circ$ : Enthalpy of vaporization at standard conditions (kJ/mol).

$logP_{oct/wat}$ : Octanol/Water partition coefficient .

$P_c$ : Critical Pressure (kPa).

$T_{boil}$ : Normal Boiling Point Temperature (K).

$T_c$ : Critical Temperature (K).

$T_{fus}$ : Normal melting (fusion) point (K).

$V_c$ : Critical Volume (m<sup>3</sup>/kg-mol).

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