

# Cholesteryl benzoate

<b>Other names:</b>	5-Cholesten-3«beta»-ol benzoate Cholesterol benzoate Cholest-5-en-3-ol (3«beta»)-, benzoate cholest-5-ene-3-beta-yl benzoate
<b>Inchi:</b>	InChI=1S/C34H50O2/c1-23(2)10-9-11-24(3)29-16-17-30-28-15-14-26-22-27(36-32(35)25
<b>InchiKey:</b>	UVZUFUGNHDDLRLQ-QGRFRJTBSA-N
<b>Formula:</b>	C34H50O2
<b>SMILES:</b>	CC(C)CCCC(C)C1CCC2C3CC=C4CC(OC(=O)c5ccccc5)CCC4(C)C3CCC12C
<b>Mol. weight [g/mol]:</b>	490.76
<b>CAS:</b>	604-32-0

## Physical Properties

Property code	Value	Unit	Source
gf	277.73	kJ/mol	Joback Method
hf	-487.75	kJ/mol	Joback Method
hfus	47.09	kJ/mol	Joback Method
hvap	100.17	kJ/mol	Joback Method
log10ws	-10.21		Crippen Method
logp	9.253		Crippen Method
mcvol	425.860	ml/mol	McGowan Method
pc	849.00	kPa	Joback Method
tb	1118.33	K	Joback Method
tc	1371.90	K	Joback Method
tf	644.04	K	Joback Method
vc	1.611	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1700.28	J/molxK	1118.33	Joback Method
cpg	1741.66	J/molxK	1160.59	Joback Method
cpg	1785.20	J/molxK	1202.85	Joback Method
cpg	1831.38	J/molxK	1245.12	Joback Method
cpg	1880.70	J/molxK	1287.38	Joback Method

cpg	1933.65	J/mol×K	1329.64	Joback Method
cpg	1990.74	J/mol×K	1371.90	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C604320&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C604320&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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