

# Butanoic acid, 4-chloro, (E)-3-hexenyl ester

<b>Inchi:</b>	InChI=1S/C10H17ClO2/c1-2-3-4-5-9-13-10(12)7-6-8-11/h3-4H,2,5-9H2,1H3/b4-3+
<b>InchiKey:</b>	JAWDGGOEFAQDOMJ-ONEGZZNKSA-N
<b>Formula:</b>	C10H17ClO2
<b>SMILES:</b>	CCC=CCCOC(=O)CCCCI
<b>Mol. weight [g/mol]:</b>	204.69

## Physical Properties

Property code	Value	Unit	Source
gf	-132.31	kJ/mol	Joback Method
hf	-393.05	kJ/mol	Joback Method
hfus	28.84	kJ/mol	Joback Method
hvap	51.35	kJ/mol	Joback Method
log10ws	-2.88		Crippen Method
logp	2.905		Crippen Method
mcvol	167.140	ml/mol	McGowan Method
pc	2231.30	kPa	Joback Method
ripol	1377.00		NIST Webbook
ripol	1397.00		NIST Webbook
ripol	1402.00		NIST Webbook
ripol	1385.00		NIST Webbook
ripol	1385.00		NIST Webbook
ripol	1875.00		NIST Webbook
ripol	1875.00		NIST Webbook
ripol	1924.00		NIST Webbook
ripol	1935.00		NIST Webbook
ripol	1960.00		NIST Webbook
tb	546.08	K	Joback Method
tc	730.87	K	Joback Method
tf	299.46	K	Joback Method
vc	0.648	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	379.88	J/mol×K	546.08	Joback Method
cpg	440.01	J/mol×K	700.07	Joback Method
cpg	429.13	J/mol×K	669.27	Joback Method
cpg	417.70	J/mol×K	638.48	Joback Method
cpg	405.69	J/mol×K	607.68	Joback Method
cpg	393.09	J/mol×K	576.88	Joback Method
cpg	450.34	J/mol×K	730.87	Joback Method
dvisc	0.0001859	Paxs	546.08	Joback Method
dvisc	0.0002420	Paxs	504.98	Joback Method
dvisc	0.0003300	Paxs	463.87	Joback Method
dvisc	0.0004780	Paxs	422.77	Joback Method
dvisc	0.0007500	Paxs	381.67	Joback Method
dvisc	0.0013117	Paxs	340.56	Joback Method
dvisc	0.0026747	Paxs	299.46	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R28988&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R28988&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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