

Glutaric acid, isobutyl 2,3,5-trichlorophenyl ester

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|----------------------|---|
| Inchi: | InChI=1S/C15H17Cl3O4/c1-9(2)8-21-13(19)4-3-5-14(20)22-12-7-10(16)6-11(17)15(12)18 |
| InchiKey: | OFLLMUNANMNDH-UHFFFAOYSA-N |
| Formula: | C15H17Cl3O4 |
| SMILES: | CC(C)COC(=O)CCCC(=O)Oc1cc(Cl)cc(Cl)c1Cl |
| Mol. weight [g/mol]: | 367.65 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -347.13 | kJ/mol | Joback Method |
| hf | -692.91 | kJ/mol | Joback Method |
| hfus | 42.12 | kJ/mol | Joback Method |
| hvap | 84.33 | kJ/mol | Joback Method |
| log10ws | -5.39 | | Crippen Method |
| logp | 4.922 | | Crippen Method |
| mcvol | 250.050 | ml/mol | McGowan Method |
| pc | 1760.97 | kPa | Joback Method |
| rinpola | 2433.00 | | NIST Webbook |
| tb | 848.65 | K | Joback Method |
| tc | 1067.46 | K | Joback Method |
| tf | 541.87 | K | Joback Method |
| vc | 0.957 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 665.23 | J/molxK | 848.65 | Joback Method |
| cpg | 676.64 | J/molxK | 885.12 | Joback Method |
| cpg | 687.02 | J/molxK | 921.59 | Joback Method |
| cpg | 696.40 | J/molxK | 958.05 | Joback Method |
| cpg | 704.76 | J/molxK | 994.52 | Joback Method |
| cpg | 712.13 | J/molxK | 1030.99 | Joback Method |
| cpg | 718.50 | J/molxK | 1067.46 | Joback Method |
| dvisc | 0.0004634 | Paxs | 541.87 | Joback Method |
| dvisc | 0.0002925 | Paxs | 593.00 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0001987 | Paxs | 644.13 | Joback Method |
| dvisc | 0.0001428 | Paxs | 695.26 | Joback Method |
| dvisc | 0.0001074 | Paxs | 746.39 | Joback Method |
| dvisc | 0.0000838 | Paxs | 797.52 | Joback Method |
| dvisc | 0.0000674 | Paxs | 848.65 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U359119&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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