

# CF<sub>3</sub>C(O)O(CH<sub>2</sub>)<sub>2</sub>CH<sub>3</sub>

<b>Other names:</b>	Acetic acid, trifluoro-, propyl ester Propyl trifluoroacetate CF <sub>3</sub> CO <sub>2</sub> (n-C <sub>3</sub> H <sub>7</sub> )
<b>Inchi:</b>	InChI=1S/C <sub>5</sub> H <sub>7</sub> F <sub>3</sub> O <sub>2</sub> /c1-2-3-10-4(9)5(6,7)8/h2-3H <sub>2</sub> ,1H <sub>3</sub>
<b>InchiKey:</b>	CDXJNCAVPFGVNL-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>5</sub> H <sub>7</sub> F <sub>3</sub> O <sub>2</sub>
<b>SMILES:</b>	CCCOC(=O)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	156.10
<b>CAS:</b>	383-66-4

## Physical Properties

Property code	Value	Unit	Source
affp	763.90	kJ/mol	NIST Webbook
basg	732.90	kJ/mol	NIST Webbook
gf	-824.29	kJ/mol	Joback Method
hf	-988.41	kJ/mol	Joback Method
hfus	13.32	kJ/mol	Joback Method
hvap	32.13	kJ/mol	Joback Method
log10ws	-1.44		Crippen Method
logp	1.502		Crippen Method
mcvol	94.060	ml/mol	McGowan Method
pc	3199.16	kPa	Joback Method
rinpol	559.00		NIST Webbook
rinpol	558.60		NIST Webbook
rinpol	569.60		NIST Webbook
tb	384.67	K	Joback Method
tc	546.37	K	Joback Method
tf	222.46	K	Joback Method
vc	0.383	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	190.04	J/molxK	384.67	Joback Method

cpg	198.49	J/mol×K	411.62	Joback Method
cpg	206.57	J/mol×K	438.57	Joback Method
cpg	214.30	J/mol×K	465.52	Joback Method
cpg	221.68	J/mol×K	492.47	Joback Method
cpg	228.73	J/mol×K	519.42	Joback Method
cpg	235.45	J/mol×K	546.37	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C383664&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C383664&amp;Units=SI</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/26-063-3/CF3C-O-O-CH2-2CH3.pdf>

Generated by Cheméo on 2024-04-20 12:57:13.850183507 +0000 UTC m=+15907082.770760822.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.