

# Acetaldehyde, tetramer

<b>Other names:</b>	1,3,5,7-Tetroxocane, 2,4,6,8-tetramethyl-Metaldehyde Ariotox Cekumeta Corry'S slug death Halizan Metacetaldehyde Metaldehyd Metaldeide Metason Slug-tox 2,4,6,8-Tetramethyl-1,3,5,7-tetroxocane UN 1332 2,4,6,8-Tetramethyl-1,3,5,7-tetraoxacyclooctane Farmon mini slug pellets R-2,C-4,C-6,C-8-tetramethyl-1,3,5,7-tetroxocane Slugfest
<b>Inchi:</b>	InChI=1S/C8H16O4/c1-5-9-6(2)11-8(4)12-7(3)10-5/h5-8H,1-4H3
<b>InchiKey:</b>	GKKDCARASOJPNG-UHFFFAOYSA-N
<b>Formula:</b>	C8H16O4
<b>SMILES:</b>	CC1OC(C)OC(C)OC(C)O1
<b>Mol. weight [g/mol]:</b>	176.21
<b>CAS:</b>	108-62-3

## Physical Properties

Property code	Value	Unit	Source
gf	-350.88	kJ/mol	Joback Method
hf	-755.47	kJ/mol	Joback Method
hfus	39.24	kJ/mol	Joback Method
hvap	51.29	kJ/mol	Joback Method
log10ws	-1.84		Crippen Method
logp	1.450		Crippen Method
mcvol	136.200	ml/mol	McGowan Method
pc	2871.95	kPa	Joback Method
tb	504.32	K	Joback Method
tc	717.90	K	Joback Method
tf	273.82	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	341.66	J/molxK	504.32	Joback Method
cpg	360.15	J/molxK	539.92	Joback Method
cpg	377.85	J/molxK	575.51	Joback Method
cpg	394.71	J/molxK	611.11	Joback Method
cpg	410.73	J/molxK	646.71	Joback Method
cpg	425.88	J/molxK	682.30	Joback Method
cpg	440.14	J/molxK	717.90	Joback Method
dvisc	0.0054245	Paxs	273.82	Joback Method
dvisc	0.0024388	Paxs	312.24	Joback Method
dvisc	0.0013063	Paxs	350.65	Joback Method
dvisc	0.0007916	Paxs	389.07	Joback Method
dvisc	0.0005248	Paxs	427.49	Joback Method
dvisc	0.0003724	Paxs	465.90	Joback Method
dvisc	0.0002784	Paxs	504.32	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C108623&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C108623&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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