

# Methyl glucuronide, acetate

<b>Inchi:</b>	InChI=1S/C13H18O10/c1-5(14)20-8-9(21-6(2)15)11(22-7(3)16)13(19-4)23-10(8)12(17)18
<b>InchiKey:</b>	YAYYBTSPWBBXRC-UHFFFAOYSA-N
<b>Formula:</b>	C13H18O10
<b>SMILES:</b>	COC1OC(C(=O)O)C(OC(C)=O)C(OC(C)=O)C1OC(C)=O
<b>Mol. weight [g/mol]:</b>	334.28

## Physical Properties

Property code	Value	Unit	Source
gf	-1106.43	kJ/mol	Joback Method
hf	-1602.12	kJ/mol	Joback Method
hfus	48.76	kJ/mol	Joback Method
hvap	101.54	kJ/mol	Joback Method
log10ws	-0.07		Crippen Method
logp	-0.763		Crippen Method
mcvol	224.670	ml/mol	McGowan Method
pc	2212.45	kPa	Joback Method
rinpol	1726.00		NIST Webbook
rinpol	1726.00		NIST Webbook
tb	922.00	K	Joback Method
tc	1134.12	K	Joback Method
tf	602.72	K	Joback Method
vc	0.829	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	750.47	J/molxK	922.00	Joback Method
cpg	782.71	J/molxK	1098.77	Joback Method
cpg	779.77	J/molxK	1063.42	Joback Method
cpg	775.01	J/molxK	1028.06	Joback Method
cpg	768.51	J/molxK	992.71	Joback Method
cpg	760.31	J/molxK	957.35	Joback Method
cpg	783.80	J/molxK	1134.12	Joback Method
dvisc	0.0000267	Paxs	922.00	Joback Method

dvisc	0.0000350	Paxs	868.79	Joback Method
dvisc	0.0000476	Paxs	815.57	Joback Method
dvisc	0.0000675	Paxs	762.36	Joback Method
dvisc	0.0001009	Paxs	709.15	Joback Method
dvisc	0.0001611	Paxs	655.93	Joback Method
dvisc	0.0002793	Paxs	602.72	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R554602&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R554602&amp;Units=SI</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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