

# Stigmasta-7,25-dien-3-ol, (3«beta»,5«alpha»)-

<b>Other names:</b>	5«alpha»-Stigmasta-7,25-dien-3«beta»-ol «delta»7,25-Stigmastadienol 24-Ethyl-5-«alpha»-cholest-7,25-dien-3-«beta»-ol 24«beta»-Ethyl-5«alpha»-cholesta-7,25-dien-3«beta»-ol Stigmasta-7,25(27)-dien-3-ol, (3«beta»,5«alpha»)-
<b>Inchi:</b>	InChI=1S/C29H48O/c1-7-21(19(2)3)9-8-20(4)25-12-13-26-24-11-10-22-18-23(30)14-16-2
<b>InchiKey:</b>	CMQUSRGUOMCHOZ-UHFFFAOYSA-N
<b>Formula:</b>	C29H48O
<b>SMILES:</b>	C=C(C)C(CC)CCC(C)C1CCC2C3=CCC4CC(O)CCC4(C)C3CCC21C
<b>Mol. weight [g/mol]:</b>	412.69
<b>CAS:</b>	6785-58-6

## Physical Properties

Property code	Value	Unit	Source
gf	299.61	kJ/mol	Joback Method
hf	-412.87	kJ/mol	Joback Method
h <sub>fus</sub>	38.81	kJ/mol	Joback Method
h <sub>vap</sub>	93.70	kJ/mol	Joback Method
log <sub>10</sub> w <sub>s</sub>	-8.69		Crippen Method
log <sub>p</sub>	7.945		Crippen Method
m <sub>cvol</sub>	373.300	ml/mol	McGowan Method
pc	982.69	kPa	Joback Method
r <sub>inpol</sub>	3325.00		NIST Webbook
tb	989.70	K	Joback Method
tc	1215.93	K	Joback Method
tf	534.21	K	Joback Method
vc	1.415	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	1425.71	J/mol×K	989.70	Joback Method
c <sub>pg</sub>	1458.39	J/mol×K	1027.41	Joback Method
c <sub>pg</sub>	1491.94	J/mol×K	1065.11	Joback Method

cpg	1526.71	J/mol×K	1102.82	Joback Method
cpg	1563.07	J/mol×K	1140.52	Joback Method
cpg	1601.37	J/mol×K	1178.23	Joback Method
cpg	1641.95	J/mol×K	1215.93	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6785586&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6785586&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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