

# Dichloroacetic acid, 4-benzyloxyphenyl ester

<b>Inchi:</b>	InChI=1S/C15H12Cl2O3/c16-14(17)15(18)20-13-8-6-12(7-9-13)19-10-11-4-2-1-3-5-11/h
<b>InchiKey:</b>	BHDHMFXXSIYFMCZ-UHFFFAOYSA-N
<b>Formula:</b>	C15H12Cl2O3
<b>SMILES:</b>	O=C(Oc1ccc(OCc2ccccc2)cc1)C(Cl)Cl
<b>Mol. weight [g/mol]:</b>	311.16

## Physical Properties

Property code	Value	Unit	Source
gf	-74.61	kJ/mol	Joback Method
hf	-305.12	kJ/mol	Joback Method
hfus	31.14	kJ/mol	Joback Method
hvap	74.15	kJ/mol	Joback Method
log10ws	-4.93		Crippen Method
logp	3.975		Crippen Method
mvol	212.480	ml/mol	McGowan Method
pc	2384.19	kPa	Joback Method
rinpol	2260.00		NIST Webbook
rinpol	2260.00		NIST Webbook
tb	774.07	K	Joback Method
tc	1018.23	K	Joback Method
tf	463.40	K	Joback Method
vc	0.793	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	540.32	J/molxK	774.07	Joback Method
cpg	592.19	J/molxK	977.54	Joback Method
cpg	584.08	J/molxK	936.84	Joback Method
cpg	574.88	J/molxK	896.15	Joback Method
cpg	564.55	J/molxK	855.46	Joback Method
cpg	553.04	J/molxK	814.76	Joback Method
cpg	599.23	J/molxK	1018.23	Joback Method
dvisc	0.0000799	Paxs	774.07	Joback Method

dvisc	0.0001017	Paxs	722.29	Joback Method
dvisc	0.0001345	Paxs	670.51	Joback Method
dvisc	0.0001862	Paxs	618.74	Joback Method
dvisc	0.0002737	Paxs	566.96	Joback Method
dvisc	0.0004347	Paxs	515.18	Joback Method
dvisc	0.0007655	Paxs	463.40	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307591&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307591&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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