

# Phenol, 2-cyclohexyl-

<b>Other names:</b>	Phenol, o-cyclohexyl- o-Cyclohexylphenol 2-Cyclohexylphenol
<b>Inchi:</b>	InChI=1S/C12H16O/c13-12-9-5-4-8-11(12)10-6-2-1-3-7-10/h4-5,8-10,13H,1-3,6-7H2
<b>InchiKey:</b>	MVRPPTGLVPEMPI-UHFFFAOYSA-N
<b>Formula:</b>	C12H16O
<b>SMILES:</b>	Oc1cccc1C1CCCCC1
<b>Mol. weight [g/mol]:</b>	176.25
<b>CAS:</b>	119-42-6

## Physical Properties

Property code	Value	Unit	Source
gf	32.40	kJ/mol	Joback Method
hf	-177.47	kJ/mol	Joback Method
hfus	18.50	kJ/mol	Joback Method
hvap	58.03	kJ/mol	Joback Method
log10ws	-3.36		Crippen Method
logp	3.440		Crippen Method
mcvol	151.190	ml/mol	McGowan Method
pc	3522.09	kPa	Joback Method
tb	600.81	K	Joback Method
tc	852.64	K	Joback Method
tf	327.90 ± 0.70	K	NIST Webbook
vc	0.498	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	394.52	J/mol×K	600.81	Joback Method
cpg	413.37	J/mol×K	642.78	Joback Method
cpg	430.73	J/mol×K	684.75	Joback Method
cpg	446.75	J/mol×K	726.72	Joback Method
cpg	461.56	J/mol×K	768.70	Joback Method
cpg	475.28	J/mol×K	810.67	Joback Method

cpg	488.07	J/mol×K	852.64	Joback Method
dvisc	0.0025484	Paxs	370.52	Joback Method
dvisc	0.0008995	Paxs	408.90	Joback Method
dvisc	0.0003796	Paxs	447.28	Joback Method
dvisc	0.0001836	Paxs	485.66	Joback Method
dvisc	0.0000988	Paxs	524.05	Joback Method
dvisc	0.0000578	Paxs	562.43	Joback Method
dvisc	0.0000363	Paxs	600.81	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C119426&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C119426&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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