

2-Pyrazoline, 5-ethyl-5-methyl

Inchi:	InChI=1S/C6H12N2/c1-3-6(2)4-5-7-8-6/h5,8H,3-4H2,1-2H3
InchiKey:	PYMHAFLRPRPQQA-UHFFFAOYSA-N
Formula:	C6H12N2
SMILES:	CCC1(C)CC=NN1
Mol. weight [g/mol]:	112.17

Physical Properties

Property code	Value	Unit	Source
gf	265.15	kJ/mol	Joback Method
hf	75.11	kJ/mol	Joback Method
hfus	14.88	kJ/mol	Joback Method
hvap	41.32	kJ/mol	Joback Method
log10ws	-1.68		Crippen Method
logp	1.134		Crippen Method
mcvol	100.200	ml/mol	McGowan Method
pc	4403.25	kPa	Joback Method
rinpol	905.00		NIST Webbook
rinpol	905.00		NIST Webbook
tb	453.61	K	Joback Method
tc	683.63	K	Joback Method
tf	369.51	K	Joback Method
vc	0.383	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	215.98	J/mol×K	453.61	Joback Method
cpg	231.26	J/mol×K	491.95	Joback Method
cpg	245.48	J/mol×K	530.28	Joback Method
cpg	258.74	J/mol×K	568.62	Joback Method
cpg	271.16	J/mol×K	606.95	Joback Method
cpg	282.83	J/mol×K	645.29	Joback Method
cpg	293.87	J/mol×K	683.63	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R511190&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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