

Fumaric acid, pentyl 2-propylphenyl ester

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| Inchi: | InChI=1S/C18H24O4/c1-3-5-8-14-21-17(19)12-13-18(20)22-16-11-7-6-10-15(16)9-4-2/h |
| InchiKey: | PUUUUMCDWXMZSN-OUKQBFOZSA-N |
| Formula: | C18H24O4 |
| SMILES: | CCCCCOC(=O)C=CC(=O)Oc1ccccc1CCC |
| Mol. weight [g/mol]: | 304.38 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -184.16 | kJ/mol | Joback Method |
| hf | -562.17 | kJ/mol | Joback Method |
| hfus | 41.80 | kJ/mol | Joback Method |
| hvap | 76.87 | kJ/mol | Joback Method |
| log10ws | -4.65 | | Crippen Method |
| logp | 3.834 | | Crippen Method |
| mcvol | 251.300 | ml/mol | McGowan Method |
| pc | 1618.07 | kPa | Joback Method |
| rinsol | 2160.00 | | NIST Webbook |
| tb | 799.64 | K | Joback Method |
| tc | 1004.52 | K | Joback Method |
| tf | 470.80 | K | Joback Method |
| vc | 0.964 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 735.73 | J/molxK | 799.64 | Joback Method |
| cpg | 750.91 | J/molxK | 833.79 | Joback Method |
| cpg | 765.08 | J/molxK | 867.93 | Joback Method |
| cpg | 778.26 | J/molxK | 902.08 | Joback Method |
| cpg | 790.48 | J/molxK | 936.23 | Joback Method |
| cpg | 801.78 | J/molxK | 970.37 | Joback Method |
| cpg | 812.18 | J/molxK | 1004.52 | Joback Method |
| dvisc | 0.0006620 | Paxs | 470.80 | Joback Method |
| dvisc | 0.0003640 | Paxs | 525.61 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0002241 | Paxs | 580.41 | Joback Method |
| dvisc | 0.0001500 | Paxs | 635.22 | Joback Method |
| dvisc | 0.0001070 | Paxs | 690.03 | Joback Method |
| dvisc | 0.0000803 | Paxs | 744.83 | Joback Method |
| dvisc | 0.0000626 | Paxs | 799.64 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U348127&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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