

n-Butyric acid 2-ethylhexyl ester

Other names:	Butanoic acid, 2-ethylhexyl ester 2-Ethylhexyl butyrate
Inchi:	InChI=1S/C12H24O2/c1-4-7-9-11(6-3)10-14-12(13)8-5-2/h11H,4-10H2,1-3H3
InchiKey:	LFEQNZNCKDNGRM-UHFFFAOYSA-N
Formula:	C12H24O2
SMILES:	CCCCC(CC)COC(=O)CCC
Mol. weight [g/mol]:	200.32
CAS:	25415-84-3

Physical Properties

Property code	Value	Unit	Source
gf	-186.20	kJ/mol	Joback Method
hf	-541.09	kJ/mol	Joback Method
hfus	26.10	kJ/mol	Joback Method
hvap	51.07	kJ/mol	Joback Method
log10ws	-3.47		Crippen Method
logp	3.546		Crippen Method
mcvol	187.380	ml/mol	McGowan Method
pc	1864.33	kPa	Joback Method
rinpol	1263.00		NIST Webbook
rinpol	1317.00		NIST Webbook
rinpol	1273.00		NIST Webbook
rinpol	1273.00		NIST Webbook
rinpol	1314.00		NIST Webbook
rinpol	1317.00		NIST Webbook
rinpol	1321.00		NIST Webbook
tb	549.81	K	Joback Method
tc	722.69	K	Joback Method
tf	282.16	K	Joback Method
vc	0.726	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	464.44	J/mol×K	549.81	Joback Method
cpg	480.46	J/mol×K	578.62	Joback Method
cpg	495.83	J/mol×K	607.44	Joback Method
cpg	510.57	J/mol×K	636.25	Joback Method
cpg	524.69	J/mol×K	665.07	Joback Method
cpg	538.20	J/mol×K	693.88	Joback Method
cpg	551.11	J/mol×K	722.69	Joback Method
dvisc	0.0042661	Pa×s	282.16	Joback Method
dvisc	0.0017460	Pa×s	326.77	Joback Method
dvisc	0.0008857	Pa×s	371.38	Joback Method
dvisc	0.0005197	Pa×s	415.99	Joback Method
dvisc	0.0003381	Pa×s	460.59	Joback Method
dvisc	0.0002373	Pa×s	505.20	Joback Method
dvisc	0.0001764	Pa×s	549.81	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C25415843&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature

tf: Normal melting (fusion) point
vc: Critical Volume

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