

# Ammonium sulfamate

<b>Other names:</b>	Ammonium amidosulfonate Sulfamic acid ammonium salt Sulfamic acid, monoammonium salt Amcide Amicide Ammat Ammate Ammate X Ammonium amidosulphate Ammonium sulphamate Ammonium sulphamidate Ammoniumsalz der amidosulfonsaure AMS Ikurin Monoammonium sulfamate Sulfamate Sulfamate dammonium Sulfaminsaure Ammate X-NI Ammonia sulfamate Ammonium amidosulfate Atlacide Necco fire retardant 2750, 2578, 2762 Root-out Sepimate Silvacide
<b>Inchi:</b>	InChI=1S/H6N2O3S/c1-5-6(2,3)4/h1H4,(H2,2,3,4)
<b>InchiKey:</b>	GWNMSRRWAFPARP-UHFFFAOYSA-N
<b>Formula:</b>	H6N2O3S
<b>SMILES:</b>	NOS(N)(=O)=O
<b>Mol. weight [g/mol]:</b>	114.12
<b>CAS:</b>	7773-06-0

## Physical Properties

Property code	Value	Unit	Source
gf	-491.52	kJ/mol	Joback Method

hf	-561.32	kJ/mol	Joback Method
h <sub>fus</sub>	18.72	kJ/mol	Joback Method
h <sub>vap</sub>	57.92	kJ/mol	Joback Method
log <sub>10</sub> w <sub>s</sub>	0.42		Crippen Method
log <sub>p</sub>	-1.920		Crippen Method
m <sub>cvol</sub>	64.780	ml/mol	McGowan Method
pc	10716.32	kPa	Joback Method
tb	414.66	K	Joback Method
tc	614.33	K	Joback Method
tf	317.07	K	Joback Method
vc	0.237	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	127.13	J/mol×K	414.66	Joback Method
cpg	131.66	J/mol×K	447.94	Joback Method
cpg	136.12	J/mol×K	481.22	Joback Method
cpg	140.50	J/mol×K	514.50	Joback Method
cpg	144.75	J/mol×K	547.77	Joback Method
cpg	148.87	J/mol×K	581.05	Joback Method
cpg	152.83	J/mol×K	614.33	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7773060&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7773060&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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