

2,5-Di(trifluoromethyl)benzoic acid, 4-nitrophenyl ester

Inchi:	InChI=1S/C15H7F6NO4/c16-14(17,18)8-1-6-12(15(19,20)21)11(7-8)13(23)26-10-4-2-9(3)
InchiKey:	HQUDURDOGOMJHN-UHFFFAOYSA-N
Formula:	C15H7F6NO4
SMILES:	O=C(Oc1ccc([N+](=O)[O-])cc1)c1cc(C(F)(F)F)ccc1C(F)(F)F
Mol. weight [g/mol]:	379.21

Physical Properties

Property code	Value	Unit	Source
gf	-1090.20	kJ/mol	Joback Method
hf	-1364.00	kJ/mol	Joback Method
hfus	39.32	kJ/mol	Joback Method
hvap	73.78	kJ/mol	Joback Method
log10ws	-6.42		Crippen Method
logp	4.852		Crippen Method
mcvol	210.170	ml/mol	McGowan Method
pc	2038.23	kPa	Joback Method
rinpol	1966.00		NIST Webbook
rinpol	1966.00		NIST Webbook
tb	828.19	K	Joback Method
tc	1053.48	K	Joback Method
tf	573.36	K	Joback Method
vc	0.852	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	609.98	J/molxK	828.19	Joback Method
cpg	619.38	J/molxK	865.74	Joback Method
cpg	627.87	J/molxK	903.29	Joback Method
cpg	635.53	J/molxK	940.84	Joback Method
cpg	642.45	J/molxK	978.38	Joback Method
cpg	648.72	J/molxK	1015.93	Joback Method
cpg	654.42	J/molxK	1053.48	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U357749&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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