

# 2,6-Octadienoic acid, 3,7-dimethyl-, methyl ester, (Z)-

<b>Other names:</b>	cis-Geranic acid methyl ester cis-2,6-Octadienoic Acid, 3,7-dimethyl-, methyl ester Methyl nerolate Z-Methyl geranate (Z)-Geranic acid, methyl ester methyl (Z)-3,7-dimethylocta-2,6-dienoate
<b>Inchi:</b>	InChI=1S/C11H18O2/c1-9(2)6-5-7-10(3)8-11(12)13-4/h6,8H,5,7H2,1-4H3/b10-8-
<b>InchiKey:</b>	ACOBBFVLNKYODD-NTMALXAHS-A-N
<b>Formula:</b>	C11H18O2
<b>SMILES:</b>	COC(=O)C=C(C)CCC=C(C)C
<b>Mol. weight [g/mol]:</b>	182.26
<b>CAS:</b>	1862-61-9

## Physical Properties

Property code	Value	Unit	Source
gf	-48.84	kJ/mol	Joback Method
hf	-300.31	kJ/mol	Joback Method
hfus	24.82	kJ/mol	Joback Method
hvap	49.31	kJ/mol	Joback Method
log10ws	-3.00		Crippen Method
logp	2.852		Crippen Method
mcvol	164.690	ml/mol	McGowan Method
pc	2237.64	kPa	Joback Method
rinpol	1279.00		NIST Webbook
rinpol	1283.00		NIST Webbook
rinpol	1302.00		NIST Webbook
rinpol	1259.00		NIST Webbook
rinpol	1298.00		NIST Webbook
rinpol	1279.00		NIST Webbook
rinpol	1280.00		NIST Webbook
ripol	1657.00		NIST Webbook
tb	535.45	K	Joback Method
tc	727.53	K	Joback Method
tf	247.81	K	Joback Method
vc	0.637	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	378.86	J/mol×K	535.45	Joback Method
cpg	393.51	J/mol×K	567.46	Joback Method
cpg	407.43	J/mol×K	599.48	Joback Method
cpg	420.66	J/mol×K	631.49	Joback Method
cpg	433.22	J/mol×K	663.51	Joback Method
cpg	445.15	J/mol×K	695.52	Joback Method
cpg	456.47	J/mol×K	727.53	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1862619&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1862619&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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