

# 1,3-Isobenzofurandione, 3a,4,7,7a-tetrahydro-

<b>Other names:</b>	4-Cyclohexene-1,2-dicarboxylic anhydride «DELTA»4-Tetrahydrophthalic anhydride Butadiene-maleic anhydride adduct Maleic anhydride adduct of butadiene Tetrahydrophthalic acid anhydride Tetrahydrophthalic anhydride 1,2,3,6-Tetrahydrophthalic acid anhydride 1,2,3,6-Tetrahydrophthalic anhydride 4-Cyclohexene-1,2-dicarboxylic acid anhydride Anhydrid kyseliny tetrahydroftalove Memtetrahydro phtalic anhydride Phthalic anhydride, 1,2,3,6-tetrahydro- Tetrahydroftalanhydrid THPA Cyclohex-4-en-1,2-dicarboxylic acid anhydride NSC 82642 Rikacid TH
<b>Inchi:</b>	InChI=1S/C8H8O3/c9-7-5-3-1-2-4-6(5)8(10)11-7/h1-2,5-6H,3-4H2
<b>InchiKey:</b>	KMOUUZVZFBCRAM-UHFFFAOYSA-N
<b>Formula:</b>	C8H8O3
<b>SMILES:</b>	O=C1OC(=O)C2CC=CCC12
<b>Mol. weight [g/mol]:</b>	152.15
<b>CAS:</b>	85-43-8

## Physical Properties

Property code	Value	Unit	Source
gf	-199.66	kJ/mol	Joback Method
hf	-430.95	kJ/mol	Joback Method
hfus	14.67	kJ/mol	Joback Method
hvap	47.04	kJ/mol	Joback Method
log10ws	-0.97		Crippen Method
logp	0.652		Crippen Method
mcvol	106.570	ml/mol	McGowan Method
pc	4140.93	kPa	Joback Method
tb	570.48	K	Joback Method
tc	825.93	K	Joback Method
tf	369.01	K	Joback Method

vc

0.395

m<sup>3</sup>/kmol

Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	272.11	J/mol×K	570.48	Joback Method
cpg	287.75	J/mol×K	613.06	Joback Method
cpg	302.42	J/mol×K	655.63	Joback Method
cpg	316.11	J/mol×K	698.21	Joback Method
cpg	328.79	J/mol×K	740.78	Joback Method
cpg	340.45	J/mol×K	783.36	Joback Method
cpg	351.08	J/mol×K	825.93	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C85438&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/26-593-5/1-3-Isobenzofurandione-3a-4-7-7a-tetrahydro.pdf>

Generated by Cheméo on 2024-04-19 14:28:13.868053211 +0000 UTC m=+15826142.788630528.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.