

# 1-Propene, 1,2,3,3-tetrachloro-

<b>Other names:</b>	1,1,2,3-Tetrachloro-2-propene 1,2,3,3-Tetrachloro-1-propene 1,2,3,3-Tetrachloropropene Propene, 1,2,3,3-tetrachloro-
<b>Inchi:</b>	InChI=1S/C3H2Cl4/c4-1-2(5)3(6)7/h1,3H/b2-1-
<b>InchiKey:</b>	JUGQRTGGLWOBPG-UPHRSURJSA-N
<b>Formula:</b>	C3H2Cl4
<b>SMILES:</b>	ClC=C(Cl)C(Cl)Cl
<b>Mol. weight [g/mol]:</b>	179.86
<b>CAS:</b>	20589-85-9

## Physical Properties

Property code	Value	Unit	Source
gf	-4.11	kJ/mol	Joback Method
hf	-66.06	kJ/mol	Joback Method
hfus	15.68	kJ/mol	Joback Method
hvap	39.46	kJ/mol	Joback Method
log10ws	-3.14		Crippen Method
logp	3.109		Crippen Method
mcvol	97.790	ml/mol	McGowan Method
pc	3960.52	kPa	Joback Method
tb	421.36	K	Joback Method
tc	641.14	K	Joback Method
tf	209.21	K	Joback Method
vc	0.374	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	133.27	J/mol×K	421.36	Joback Method
cpg	138.11	J/mol×K	457.99	Joback Method
cpg	142.53	J/mol×K	494.62	Joback Method
cpg	146.55	J/mol×K	531.25	Joback Method
cpg	150.20	J/mol×K	567.88	Joback Method

cpg	153.52	J/mol×K	604.51	Joback Method
cpg	156.54	J/mol×K	641.14	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.62835e+01
Coeff. B	-4.11091e+03
Coeff. C	-6.21220e+01
Temperature range (K), min.	319.12
Temperature range (K), max.	436.80

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20589859&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20589859&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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