

# N,N-DIMETHYLOCTADECYLAMINE

## Other names:

(Hydrochloride salt) dimantine  
1-Octadecanamine, N,N-dimethyl-  
ADMA 18  
Adogen 342D  
Armeen DM 18D  
Armine DM18D  
Barlene 18S  
DM 18D  
Dimanthine  
Dimethyl stearamine  
Dimethyl-1-octadecanamine  
Dimethyl-n-octadecylamine  
Dimethyloctadecylamine  
Dimethylstearylamine  
Dymanthine  
Farmin DM 80  
Kemamine 9902D  
Kemamine T-9902  
N,N-Dimethyl-1-octadecanamine  
N,N-Dimethyl-n-octadecylamine  
N,N-Dimethyloctadecylamin  
N,N-Dimethylstearylamine  
N,N-dimethyl octadecanamine  
N-Octadecyl-N,N-dimethylamine  
N-Stearyl-N,N-dimethylamine  
Octadecylamine, N,N-dimethyl-  
Octadecyldimethylamine  
Onamine 18  
Stearyldimethylamine

## Inchi:

InChI=1S/C20H43N/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21(2)3/h4-20H2,1

## InchiKey:

NAPSCFZYZVSQHF-UHFFFAOYSA-N

## Formula:

C20H43N

## SMILES:

CCCCCCCCCCCCCCCCCN(C)C

## Mol. weight [g/mol]:

297.57

## CAS:

124-28-7

## Physical Properties

| Property code | Value         | Unit    | Source             |
|---------------|---------------|---------|--------------------|
| af            | 0.8714        |         | Cheméo Relay (1.0) |
| gf            | 228.30        | kJ/mol  | Joback Method      |
| hf            | -397.61       | kJ/mol  | Cheméo Relay (1.0) |
| hfus          | 50.58         | kJ/mol  | Joback Method      |
| hvap          | 99.75         | kJ/mol  | Cheméo Relay (1.0) |
| ie            | 8.44          | eV      | Cheméo Relay (1.0) |
| log10ws       | -7.19         |         | Cheméo Relay (1.0) |
| logp          | 6.809         |         | Crippen Method     |
| mcvol         | 302.640       | ml/mol  | McGowan Method     |
| pc            | 1014.24       | kPa     | Joback Method      |
| rinpol        | 2096.00       |         | NIST Webbook       |
| rinpol        | 2096.00       |         | NIST Webbook       |
| tb            | 591.58        | K       | Cheméo Relay (1.0) |
| tc            | 755.92        | K       | Cheméo Relay (1.0) |
| tf            | 296.04 ± 0.20 | K       | NIST Webbook       |
| tf            | 296.04 ± 0.20 | K       | NIST Webbook       |
| tf            | 301.00 ± 1.50 | K       | NIST Webbook       |
| tf            | 298.15 ± 1.00 | K       | NIST Webbook       |
| vc            | 1.215         | m3/kmol | Cheméo Relay (1.0) |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 883.36 | J/mol×K | 669.44          | Joback Method |
| cpg           | 904.53 | J/mol×K | 696.23          | Joback Method |
| cpg           | 924.80 | J/mol×K | 723.01          | Joback Method |
| cpg           | 944.20 | J/mol×K | 749.80          | Joback Method |
| cpg           | 962.77 | J/mol×K | 776.58          | Joback Method |
| cpg           | 980.54 | J/mol×K | 803.37          | Joback Method |
| cpg           | 997.52 | J/mol×K | 830.16          | Joback Method |
| hvapt         | 74.70  | kJ/mol  | 602.50          | NIST Webbook  |

## Correlations

| Information   | Value |
|---------------|-------|
| Property code | pvap  |

| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
|-----------------------------|-------------------------------|
| Coeff. A                    | 1.72220e+01                   |
| Coeff. B                    | -6.25332e+03                  |
| Coeff. C                    | -1.16054e+02                  |
| Temperature range (K), min. | 485.32                        |
| Temperature range (K), max. | 641.08                        |

## Sources

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

The Yaws Handbook of Vapor  
Pressure:  
NIST Webbook:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C124287&Units=SI>

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>af:</b>      | Acentric Factor                                 |
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>hvapt:</b>   | Enthalpy of vaporization at a given temperature |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>pvap:</b>    | Vapor pressure                                  |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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