

# Butanoic acid, 2-methyl-, 1-methylethyl ester

<b>Other names:</b>	Isopropyl 2-methylbutanoate 2-Propyl 2-methylbutanoate Isopropyl 2-methylbutyrate 1-Methylethyl 2-methylbutanoate isopropyl 2-methylbutyrate
<b>Inchi:</b>	InChI=1S/C8H16O2/c1-5-7(4)8(9)10-6(2)3/h6-7H,5H2,1-4H3
<b>InchiKey:</b>	DIRDKDDFAMNBNY-UHFFFAOYSA-N
<b>Formula:</b>	C8H16O2
<b>SMILES:</b>	CCC(C)C(=O)OC(C)C
<b>Mol. weight [g/mol]:</b>	144.21
<b>CAS:</b>	66576-71-4

## Physical Properties

Property code	Value	Unit	Source
gf	-222.32	kJ/mol	Joback Method
hf	-463.81	kJ/mol	Joback Method
hfus	12.22	kJ/mol	Joback Method
hvap	41.78	kJ/mol	Joback Method
log10ws	-1.90		Crippen Method
logp	1.984		Crippen Method
mcvol	131.020	ml/mol	McGowan Method
pc	2701.41	kPa	Joback Method
rinpol	960.00		NIST Webbook
ripol	1052.00		NIST Webbook
ripol	1061.00		NIST Webbook
tb	457.85	K	Joback Method
tc	640.65	K	Joback Method
tf	222.08	K	Joback Method
vc	0.495	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	279.95	J/molxK	457.85	Joback Method

cpg	292.89	J/mol×K	488.32	Joback Method
cpg	305.35	J/mol×K	518.78	Joback Method
cpg	317.32	J/mol×K	549.25	Joback Method
cpg	328.83	J/mol×K	579.72	Joback Method
cpg	339.86	J/mol×K	610.19	Joback Method
cpg	350.42	J/mol×K	640.65	Joback Method
dvisc	0.0076048	Paxs	222.08	Joback Method
dvisc	0.0027406	Paxs	261.38	Joback Method
dvisc	0.0012896	Paxs	300.67	Joback Method
dvisc	0.0007223	Paxs	339.97	Joback Method
dvisc	0.0004562	Paxs	379.26	Joback Method
dvisc	0.0003141	Paxs	418.56	Joback Method
dvisc	0.0002306	Paxs	457.85	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C66576714&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C66576714&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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