

# 3-Methoxy-4,5-methylenedioxybenzaldehyde

<b>Other names:</b>	1,3-Benzodioxole-5-carboxaldehyde, 7-methoxy-3-Methoxypiperonal 7-methoxybenzo-1,3-dioxole-5-carboxaldehyde
<b>Inchi:</b>	InChI=1S/C9H8O4/c1-11-7-2-6(4-10)3-8-9(7)13-5-12-8/h2-4H,5H2,1H3
<b>InchiKey:</b>	LOFRBHZYZCIOJO-UHFFFAOYSA-N
<b>Formula:</b>	C9H8O4
<b>SMILES:</b>	COc1cc(C=O)cc2c1OCO2
<b>Mol. weight [g/mol]:</b>	180.16
<b>CAS:</b>	5780-07-4

## Physical Properties

Property code	Value	Unit	Source
gf	-199.88	kJ/mol	Joback Method
hf	-415.63	kJ/mol	Joback Method
hfus	28.44	kJ/mol	Joback Method
hvap	58.26	kJ/mol	Joback Method
log10ws	-2.04		Crippen Method
logp	1.236		Crippen Method
mcvol	122.230	ml/mol	McGowan Method
pc	3935.71	kPa	Joback Method
rinpol	1570.00		NIST Webbook
rinpol	1570.00		NIST Webbook
tb	583.33	K	Joback Method
tc	809.74	K	Joback Method
tf	394.72	K	Joback Method
vc	0.467	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	293.11	J/molxK	583.33	Joback Method
cpg	303.68	J/molxK	621.06	Joback Method
cpg	313.57	J/molxK	658.80	Joback Method
cpg	322.81	J/molxK	696.53	Joback Method

cpg	331.43	J/molxK	734.27	Joback Method
cpg	339.47	J/molxK	772.00	Joback Method
cpg	346.95	J/molxK	809.74	Joback Method
dvisc	0.0016780	Paxs	394.72	Joback Method
dvisc	0.0012589	Paxs	426.16	Joback Method
dvisc	0.0009825	Paxs	457.59	Joback Method
dvisc	0.0007916	Paxs	489.02	Joback Method
dvisc	0.0006547	Paxs	520.46	Joback Method
dvisc	0.0005532	Paxs	551.89	Joback Method
dvisc	0.0004761	Paxs	583.33	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5780074&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5780074&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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