

2,6-Difluoro-3-methylbenzoic acid, 4-chlorophenyl ester

Inchi:	InChI=1S/C14H9ClF2O2/c1-8-2-7-11(16)12(13(8)17)14(18)19-10-5-3-9(15)4-6-10/h2-7H
InchiKey:	MZKJPBVMPYPFGX-UHFFFAOYSA-N
Formula:	C14H9ClF2O2
SMILES:	Cc1ccc(F)c(C(=O)Oc2ccc(Cl)cc2)c1F
Mol. weight [g/mol]:	282.67

Physical Properties

Property code	Value	Unit	Source
gf	-382.17	kJ/mol	Joback Method
hf	-557.87	kJ/mol	Joback Method
hfus	31.69	kJ/mol	Joback Method
hvap	65.86	kJ/mol	Joback Method
log10ws	-5.38		Crippen Method
logp	4.146		Crippen Method
mvol	183.820	ml/mol	McGowan Method
pc	2400.57	kPa	Joback Method
rinpol	1968.00		NIST Webbook
rinpol	1968.00		NIST Webbook
tb	705.26	K	Joback Method
tc	932.46	K	Joback Method
tf	453.72	K	Joback Method
vc	0.713	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	451.72	J/mol×K	705.26	Joback Method
cpg	463.72	J/mol×K	743.13	Joback Method
cpg	474.81	J/mol×K	780.99	Joback Method
cpg	485.03	J/mol×K	818.86	Joback Method
cpg	494.38	J/mol×K	856.73	Joback Method
cpg	502.89	J/mol×K	894.60	Joback Method
cpg	510.58	J/mol×K	932.46	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U357682&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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