

# Diethylmalonic acid, decyl 2-methoxyethyl ester

Inchi:	InChI=1S/C20H38O5/c1-5-8-9-10-11-12-13-14-15-24-18(21)20(6-2,7-3)19(22)25-17-16-2
InchiKey:	ZBUWRECEUGOTSB-UHFFFAOYSA-N
Formula:	C20H38O5
SMILES:	CCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCCOC
Mol. weight [g/mol]:	358.51

## Physical Properties

Property code	Value	Unit	Source
gf	-452.48	kJ/mol	Joback Method
hf	-1086.70	kJ/mol	Joback Method
hfus	46.90	kJ/mol	Joback Method
hvap	79.54	kJ/mol	Joback Method
log10ws	-4.77		Crippen Method
logp	4.666		Crippen Method
mcvol	313.410	ml/mol	McGowan Method
pc	1079.22	kPa	Joback Method
rinpol	2183.00		NIST Webbook
rinpol	2183.00		NIST Webbook
tb	828.77	K	Joback Method
tc	1017.43	K	Joback Method
tf	484.13	K	Joback Method
vc	1.210	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1001.66	J/molxK	828.77	Joback Method
cpg	1019.60	J/molxK	860.21	Joback Method
cpg	1036.43	J/molxK	891.66	Joback Method
cpg	1052.17	J/molxK	923.10	Joback Method
cpg	1066.82	J/molxK	954.55	Joback Method
cpg	1080.43	J/molxK	985.99	Joback Method
cpg	1093.00	J/molxK	1017.43	Joback Method
dvisc	0.0005354	Paxs	484.13	Joback Method

dvisc	0.0002603	Paxs	541.57	Joback Method
dvisc	0.0001453	Paxs	599.01	Joback Method
dvisc	0.0000898	Paxs	656.45	Joback Method
dvisc	0.0000600	Paxs	713.89	Joback Method
dvisc	0.0000425	Paxs	771.33	Joback Method
dvisc	0.0000316	Paxs	828.77	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370678&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370678&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/26-889-7/Diethylmalonic-acid-decyl-2-methoxyethyl-ester.pdf>

Generated by Cheméo on 2024-04-26 21:06:04.451969071 +0000 UTC m=+16454813.372546384.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.