

# 1-Penten-3-one, 1-phenyl-

<b>Other names:</b>	Benzylidenemethyl ethyl ketone Ethyl styryl ketone Ethyl 2-phenylvinyl ketone Styryl ethyl ketone 1-Phenyl-1-penten-3-one 1-phenylpent-1-en-3-one
<b>Inchi:</b>	InChI=1S/C11H12O/c1-2-11(12)9-8-10-6-4-3-5-7-10/h3-9H,2H2,1H3/b9-8+
<b>InchiKey:</b>	LVGUHATVVHIJET-CMDGGOBGSA-N
<b>Formula:</b>	C11H12O
<b>SMILES:</b>	CCC(=O)C=Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	160.21
<b>CAS:</b>	3152-68-9

## Physical Properties

Property code	Value	Unit	Source
gf	105.45	kJ/mol	Joback Method
hf	-29.20	kJ/mol	Joback Method
hfus	20.09	kJ/mol	Joback Method
hvap	49.06	kJ/mol	Joback Method
log10ws	-2.83		Crippen Method
logp	2.679		Crippen Method
mcvol	139.360	ml/mol	McGowan Method
pc	2992.59	kPa	Joback Method
tb	535.79	K	Joback Method
tc	757.15	K	Joback Method
tf	285.00	K	Joback Method
vc	0.529	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	303.19	J/mol×K	535.79	Joback Method
cpg	365.53	J/mol×K	720.26	Joback Method
cpg	354.78	J/mol×K	683.36	Joback Method

cpg	343.22	J/molxK	646.47	Joback Method
cpg	330.81	J/molxK	609.58	Joback Method
cpg	317.48	J/molxK	572.68	Joback Method
cpg	375.54	J/molxK	757.15	Joback Method
dvisc	0.0002065	Paxs	535.79	Joback Method
dvisc	0.0002659	Paxs	493.99	Joback Method
dvisc	0.0003588	Paxs	452.19	Joback Method
dvisc	0.0005148	Paxs	410.39	Joback Method
dvisc	0.0008014	Paxs	368.60	Joback Method
dvisc	0.0013973	Paxs	326.80	Joback Method
dvisc	0.0028677	Paxs	285.00	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	417.00 ± 1.00	K	1.60	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3152689&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3152689&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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