

# Benzene, 1-ethoxy-2,4-dinitro-

<b>Other names:</b>	2,4-Dinitrophenetole 2,4-Dinitrophenyl ethyl ether 1-Ethoxy-2,4-dinitrobenzene Phenetole, 2,4-dinitro- 2,4-Dinitro-1-ethoxy-benzene 2,4-Dinitrofenetol
<b>Inchi:</b>	InChI=1S/C8H8N2O5/c1-2-15-8-4-3-6(9(11)12)5-7(8)10(13)14/h3-5H,2H2,1H3
<b>InchiKey:</b>	YSOKMOXAGMIZFZ-UHFFFAOYSA-N
<b>Formula:</b>	C8H8N2O5
<b>SMILES:</b>	CCOC1CC([N+](=O)[O-])CC1[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	212.16
<b>CAS:</b>	610-54-8

## Physical Properties

Property code	Value	Unit	Source
chs	-4064.10 ± 4.10	kJ/mol	NIST Webbook
chs	-4065.90 ± 4.10	kJ/mol	NIST Webbook
gf	75.73	kJ/mol	Joback Method
hf	-148.60	kJ/mol	Joback Method
hfus	33.65	kJ/mol	Joback Method
hvap	72.59	kJ/mol	Joback Method
log10ws	-3.31		Crippen Method
logp	1.902		Crippen Method
mcvol	140.530	ml/mol	McGowan Method
pc	3551.53	kPa	Joback Method
tb	745.18	K	Joback Method
tc	1007.50	K	Joback Method
tf	540.83	K	Joback Method
vc	0.557	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	371.15	J/mol×K	745.18	Joback Method

cpg	381.49	J/mol×K	788.90	Joback Method
cpg	390.88	J/mol×K	832.62	Joback Method
cpg	399.33	J/mol×K	876.34	Joback Method
cpg	406.88	J/mol×K	920.06	Joback Method
cpg	413.54	J/mol×K	963.78	Joback Method
cpg	419.35	J/mol×K	1007.50	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C610548&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C610548&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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