

# (-)-1,2:5,6-Di-O-cyclohexylidene-L-inositol, diacetate

<b>Inchi:</b>	InChI=1S/C22H32O8/c1-13(23)25-15-16(26-14(2)24)18-20(30-22(28-18)11-7-4-8-12-22)
<b>InchiKey:</b>	ZUFNLWFMJYRZHS-UHFFFAOYSA-N
<b>Formula:</b>	C22H32O8
<b>SMILES:</b>	CC(=O)OC1C(OC(C)=O)C2OC3(CCCCC3)OC2C2OC3(CCCCC3)OC12
<b>Mol. weight [g/mol]:</b>	424.48

## Physical Properties

Property code	Value	Unit	Source
gf	-493.02	kJ/mol	Joback Method
hf	-1224.67	kJ/mol	Joback Method
hfus	56.82	kJ/mol	Joback Method
hvap	98.46	kJ/mol	Joback Method
log10ws	-4.46		Crippen Method
logp	2.752		Crippen Method
mvol	304.900	ml/mol	McGowan Method
pc	1655.15	kPa	Joback Method
rinpol	2664.50		NIST Webbook
rinpol	2664.50		NIST Webbook
tb	1013.20	K	Joback Method
tc	1264.20	K	Joback Method
tf	688.44	K	Joback Method
vc	1.121	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1202.29	J/molxK	1013.20	Joback Method
cpg	1231.45	J/molxK	1055.03	Joback Method
cpg	1261.56	J/molxK	1096.87	Joback Method
cpg	1293.02	J/molxK	1138.70	Joback Method
cpg	1326.21	J/molxK	1180.53	Joback Method
cpg	1361.53	J/molxK	1222.37	Joback Method
cpg	1399.37	J/molxK	1264.20	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U380076&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U380076&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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