

# Pentanoic acid, 2-iodoethyl ester

<b>Inchi:</b>	InChI=1S/C7H13IO2/c1-2-3-4-7(9)10-6-5-8/h2-6H2,1H3
<b>InchiKey:</b>	VLKAEOLCZKBFBS-UHFFFAOYSA-N
<b>Formula:</b>	C7H13IO2
<b>SMILES:</b>	CCCCC(=O)OCCI
<b>Mol. weight [g/mol]:</b>	256.08

## Physical Properties

Property code	Value	Unit	Source
gf	-167.74	kJ/mol	Joback Method
hf	-355.74	kJ/mol	Joback Method
hfus	21.08	kJ/mol	Joback Method
hvap	49.70	kJ/mol	Joback Method
log10ws	-2.56		Crippen Method
logp	2.155		Crippen Method
mcvol	142.750	ml/mol	McGowan Method
pc	2871.95	kPa	Joback Method
rinpola	1250.00		NIST Webbook
rinpola	1250.00		NIST Webbook
tb	528.99	K	Joback Method
tc	735.45	K	Joback Method
tf	298.87	K	Joback Method
vc	0.539	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	288.01	J/molxK	528.99	Joback Method
cpg	338.23	J/molxK	701.04	Joback Method
cpg	329.21	J/molxK	666.63	Joback Method
cpg	319.70	J/molxK	632.22	Joback Method
cpg	309.66	J/molxK	597.81	Joback Method
cpg	299.10	J/molxK	563.40	Joback Method
cpg	346.75	J/molxK	735.45	Joback Method
dvisc	0.0002947	Paxs	528.99	Joback Method

dvisc	0.0003785	Paxs	490.64	Joback Method
dvisc	0.0005072	Paxs	452.28	Joback Method
dvisc	0.0007175	Paxs	413.93	Joback Method
dvisc	0.0010895	Paxs	375.58	Joback Method
dvisc	0.0018193	Paxs	337.22	Joback Method
dvisc	0.0034653	Paxs	298.87	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R20003&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R20003&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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