

# 2-Bromobenzoic acid, 2-naphthyl ester

<b>Inchi:</b>	InChI=1S/C17H11BrO2/c18-16-8-4-3-7-15(16)17(19)20-14-10-9-12-5-1-2-6-13(12)11-14
<b>InchiKey:</b>	HNTXBLZKNZXOPY-UHFFFAOYSA-N
<b>Formula:</b>	C17H11BrO2
<b>SMILES:</b>	O=C(Oc1ccc2ccccc2c1)c1ccccc1Br
<b>Mol. weight [g/mol]:</b>	327.17

## Physical Properties

Property code	Value	Unit	Source
gf	184.87	kJ/mol	Joback Method
hf	28.51	kJ/mol	Joback Method
hfus	32.18	kJ/mol	Joback Method
hvap	76.54	kJ/mol	Joback Method
log10ws	-6.53		Crippen Method
logp	4.822		Crippen Method
mvol	208.350	ml/mol	McGowan Method
pc	2896.73	kPa	Joback Method
rinpol	2520.00		NIST Webbook
rinpol	2520.00		NIST Webbook
tb	813.11	K	Joback Method
tc	1079.90	K	Joback Method
tf	523.89	K	Joback Method
vc	0.779	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	533.76	J/molxK	813.11	Joback Method
cpg	585.95	J/molxK	1035.44	Joback Method
cpg	577.22	J/molxK	990.97	Joback Method
cpg	567.77	J/molxK	946.51	Joback Method
cpg	557.45	J/molxK	902.04	Joback Method
cpg	546.16	J/molxK	857.58	Joback Method
cpg	594.09	J/molxK	1079.90	Joback Method
dvisc	0.0001834	Paxs	813.11	Joback Method

dvisc	0.0002180	Paxs	764.91	Joback Method
dvisc	0.0002653	Paxs	716.70	Joback Method
dvisc	0.0003321	Paxs	668.50	Joback Method
dvisc	0.0004304	Paxs	620.30	Joback Method
dvisc	0.0005827	Paxs	572.09	Joback Method
dvisc	0.0008342	Paxs	523.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307749&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307749&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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