

# Ethylloflazepate, hydrolysis, acetylated

**Inchi:** InChI=1S/C15H11ClFNO2/c1-9(19)18-14-7-6-10(16)8-12(14)15(20)11-4-2-3-5-13(11)17/  
**InchiKey:** BIJCUDKFSIRCDI-UHFFFAOYSA-N  
**Formula:** C15H11ClFNO2  
**SMILES:** CC(=O)Nc1ccc(Cl)cc1C(=O)c1ccccc1F  
**Mol. weight [g/mol]:** 291.70

## Physical Properties

Property code	Value	Unit	Source
gf	-103.84	kJ/mol	Joback Method
hf	-297.82	kJ/mol	Joback Method
hfus	37.09	kJ/mol	Joback Method
hvap	79.02	kJ/mol	Joback Method
log10ws	-4.65		Crippen Method
logp	3.668		Crippen Method
mcvol	201.820	ml/mol	McGowan Method
pc	2540.49	kPa	Joback Method
rinpol	2195.00		NIST Webbook
rinpol	2195.00		NIST Webbook
rinpol	2195.00		NIST Webbook
rinpol	2195.00		NIST Webbook
rinpol	2195.00		NIST Webbook
tb	805.51	K	Joback Method
tc	1043.82	K	Joback Method
tf	532.24	K	Joback Method
vc	0.773	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	531.69	J/molxK	805.51	Joback Method
cpg	542.86	J/molxK	845.23	Joback Method
cpg	553.02	J/molxK	884.95	Joback Method
cpg	562.23	J/molxK	924.67	Joback Method
cpg	570.55	J/molxK	964.39	Joback Method

cpg	578.03	J/mol×K	1004.10	Joback Method
cpg	584.72	J/mol×K	1043.82	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R313005&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R313005&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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