

# 6-methoxy-dihydroindole

<b>Inchi:</b>	InChI=1S/C9H11NO/c1-11-8-3-2-7-4-5-10-9(7)6-8/h2-3,6,10H,4-5H2,1H3
<b>InchiKey:</b>	GKFGHNMPMAXWQS-UHFFFAOYSA-N
<b>Formula:</b>	C9H11NO
<b>SMILES:</b>	COc1ccc2c(c1)NCC2
<b>Mol. weight [g/mol]:</b>	149.19

## Physical Properties

Property code	Value	Unit	Source
gf	169.22	kJ/mol	Joback Method
hf	-16.77	kJ/mol	Joback Method
hfus	20.17	kJ/mol	Joback Method
hvap	48.62	kJ/mol	Joback Method
log10ws	-1.88		Crippen Method
logp	1.663		Crippen Method
mcvol	118.900	ml/mol	McGowan Method
pc	3857.88	kPa	Joback Method
rinpol	1515.00		NIST Webbook
rinpol	1503.00		NIST Webbook
rinpol	1503.00		NIST Webbook
ripol	2384.00		NIST Webbook
ripol	2412.00		NIST Webbook
ripol	2384.00		NIST Webbook
tb	524.34	K	Joback Method
tc	756.04	K	Joback Method
tf	392.09	K	Joback Method
vc	0.445	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	262.73	J/molxK	524.34	Joback Method
cpg	276.35	J/molxK	562.96	Joback Method
cpg	289.15	J/molxK	601.57	Joback Method
cpg	301.15	J/molxK	640.19	Joback Method

cpg	312.40	J/mol×K	678.81	Joback Method
cpg	322.93	J/mol×K	717.42	Joback Method
cpg	332.78	J/mol×K	756.04	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R135236&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R135236&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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