

# 4-Chlorobutyric acid, 4-biphenyl ester

<b>Inchi:</b>	InChI=1S/C16H15ClO2/c17-12-4-7-16(18)19-15-10-8-14(9-11-15)13-5-2-1-3-6-13/h1-3,5
<b>InchiKey:</b>	CTIRJFONWYTVGL-UHFFFAOYSA-N
<b>Formula:</b>	C16H15ClO2
<b>SMILES:</b>	O=C(CCCCl)Oc1ccc(-c2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	274.74

## Physical Properties

Property code	Value	Unit	Source
gf	53.18	kJ/mol	Joback Method
hf	-172.52	kJ/mol	Joback Method
hfus	31.87	kJ/mol	Joback Method
hvap	69.97	kJ/mol	Joback Method
log10ws	-5.38		Crippen Method
logp	4.278		Crippen Method
mcvol	208.460	ml/mol	McGowan Method
pc	2271.90	kPa	Joback Method
rinsol	2355.00		NIST Webbook
tb	737.54	K	Joback Method
tc	973.82	K	Joback Method
tf	437.52	K	Joback Method
vc	0.788	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	543.75	J/molxK	737.54	Joback Method
cpg	558.38	J/molxK	776.92	Joback Method
cpg	571.82	J/molxK	816.30	Joback Method
cpg	584.13	J/molxK	855.68	Joback Method
cpg	595.37	J/molxK	895.06	Joback Method
cpg	605.58	J/molxK	934.44	Joback Method
cpg	614.84	J/molxK	973.82	Joback Method
dvisc	0.0010146	Paxs	437.52	Joback Method
dvisc	0.0005818	Paxs	487.52	Joback Method

dvisc	0.0003700	Paxs	537.53	Joback Method
dvisc	0.0002541	Paxs	587.53	Joback Method
dvisc	0.0001852	Paxs	637.53	Joback Method
dvisc	0.0001413	Paxs	687.54	Joback Method
dvisc	0.0001118	Paxs	737.54	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U360647&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U360647&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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