

# Glutaric acid, 2,5-dimethylphenyl pentyl ester

**Inchi:** InChI=1S/C18H26O4/c1-4-5-6-12-21-17(19)8-7-9-18(20)22-16-13-14(2)10-11-15(16)3/h1-18  
**InchiKey:** FOMXTHGRKKBKPHC-UHFFFAOYSA-N  
**Formula:** C18H26O4  
**SMILES:** CCCCCOC(=O)CCCC(=O)Oc1cc(C)ccc1C  
**Mol. weight [g/mol]:** 306.40

## Physical Properties

Property code	Value	Unit	Source
gf	-274.01	kJ/mol	Joback Method
hf	-690.86	kJ/mol	Joback Method
hfus	41.21	kJ/mol	Joback Method
hvap	77.57	kJ/mol	Joback Method
log10ws	-4.95		Crippen Method
logp	4.113		Crippen Method
mvol	255.600	ml/mol	McGowan Method
pc	1534.26	kPa	Joback Method
rinpol	2276.00		NIST Webbook
rinpol	2276.00		NIST Webbook
tb	800.46	K	Joback Method
tc	1000.95	K	Joback Method
tf	488.40	K	Joback Method
vc	0.984	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	760.53	J/molxK	800.46	Joback Method
cpg	828.18	J/molxK	967.54	Joback Method
cpg	816.68	J/molxK	934.12	Joback Method
cpg	804.18	J/molxK	900.71	Joback Method
cpg	790.66	J/molxK	867.29	Joback Method
cpg	776.12	J/molxK	833.88	Joback Method
cpg	838.68	J/molxK	1000.95	Joback Method
dvisc	0.0000726	Paxs	800.46	Joback Method

dvisc	0.0000916	Paxs	748.45	Joback Method
dvisc	0.0001196	Paxs	696.44	Joback Method
dvisc	0.0001630	Paxs	644.43	Joback Method
dvisc	0.0002346	Paxs	592.42	Joback Method
dvisc	0.0003622	Paxs	540.41	Joback Method
dvisc	0.0006133	Paxs	488.40	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358938&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358938&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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