

# Bicyclo[2.2.1]hept-2-ene, 5-methylene-

<b>Other names:</b>	5-Methylene-2-norbornene 2-Methylene-5-norbornene 2-Norbornene, 5-methylene- 5-Methylene-2-norbornylene 5-Methylenebicyclo(2.2.1)hept-2-ene 5-Methylenenorbornene 5-Methylenebicyclo[2.2.1]-2-heptene
<b>Inchi:</b>	InChI=1S/C8H10/c1-6-4-7-2-3-8(6)5-7/h2-3,7-8H,1,4-5H2
<b>InchiKey:</b>	WTQBISBWKRLIJ-UHFFFAOYSA-N
<b>Formula:</b>	C8H10
<b>SMILES:</b>	C=C1CC2C=CC1C2
<b>Mol. weight [g/mol]:</b>	106.17
<b>CAS:</b>	694-91-7

## Physical Properties

Property code	Value	Unit	Source
gf	208.92	kJ/mol	Joback Method
hf	73.01	kJ/mol	Joback Method
hfus	10.71	kJ/mol	Joback Method
hvap	33.85	kJ/mol	Joback Method
ie	9.01	eV	NIST Webbook
ie	8.93	eV	NIST Webbook
log10ws	-2.18		Crippen Method
logp	2.139		Crippen Method
mcvol	93.260	ml/mol	McGowan Method
pc	3690.97	kPa	Joback Method
rinpol	800.00		NIST Webbook
rinpol	803.00		NIST Webbook
rinpol	800.00		NIST Webbook
rinpol	800.00		NIST Webbook
rinpol	803.00		NIST Webbook
rinpol	798.00		NIST Webbook
tb	398.51	K	Joback Method
tc	603.88	K	Joback Method
tf	226.72	K	Joback Method
vc	0.359	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	174.97	J/molxK	398.51	Joback Method
cpg	189.30	J/molxK	432.74	Joback Method
cpg	202.67	J/molxK	466.97	Joback Method
cpg	215.16	J/molxK	501.20	Joback Method
cpg	226.80	J/molxK	535.43	Joback Method
cpg	237.67	J/molxK	569.66	Joback Method
cpg	247.80	J/molxK	603.88	Joback Method
dvisc	0.0004647	Paxs	226.72	Joback Method
dvisc	0.0004653	Paxs	255.35	Joback Method
dvisc	0.0004658	Paxs	283.98	Joback Method
dvisc	0.0004662	Paxs	312.62	Joback Method
dvisc	0.0004666	Paxs	341.25	Joback Method
dvisc	0.0004669	Paxs	369.88	Joback Method
dvisc	0.0004671	Paxs	398.51	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C694917&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C694917&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinqol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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