

# Dimethylmalonic acid, 2,5-dichlorophenyl propyl ester

<b>Inchi:</b>	InChI=1S/C14H16Cl2O4/c1-4-7-19-12(17)14(2,3)13(18)20-11-8-9(15)5-6-10(11)16/h5-6,
<b>InchiKey:</b>	NKOYRWAXUGHUPI-UHFFFAOYSA-N
<b>Formula:</b>	C14H16Cl2O4
<b>SMILES:</b>	CCCOC(=O)C(C)(C)C(=O)Oc1cc(Cl)ccc1Cl
<b>Mol. weight [g/mol]:</b>	319.18

## Physical Properties

Property code	Value	Unit	Source
gf	-328.71	kJ/mol	Joback Method
hf	-648.53	kJ/mol	Joback Method
hfus	31.83	kJ/mol	Joback Method
hvap	76.14	kJ/mol	Joback Method
log10ws	-4.29		Crippen Method
logp	3.878		Crippen Method
mcvol	223.720	ml/mol	McGowan Method
pc	2021.76	kPa	Joback Method
rinpol	1953.00		NIST Webbook
rinpol	1953.00		NIST Webbook
tb	780.57	K	Joback Method
tc	1003.61	K	Joback Method
tf	505.58	K	Joback Method
vc	0.847	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	591.06	J/molxK	780.57	Joback Method
cpg	643.09	J/molxK	966.44	Joback Method
cpg	634.55	J/molxK	929.26	Joback Method
cpg	625.11	J/molxK	892.09	Joback Method
cpg	614.74	J/molxK	854.92	Joback Method
cpg	603.39	J/molxK	817.74	Joback Method
cpg	650.75	J/molxK	1003.61	Joback Method
dvisc	0.0000775	Paxs	780.57	Joback Method

dvisc	0.0000975	Paxs	734.74	Joback Method
dvisc	0.0001266	Paxs	688.91	Joback Method
dvisc	0.0001705	Paxs	643.07	Joback Method
dvisc	0.0002404	Paxs	597.24	Joback Method
dvisc	0.0003589	Paxs	551.41	Joback Method
dvisc	0.0005761	Paxs	505.58	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U363678&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U363678&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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