

# Hydrazine, 1,1-diphenyl-

<b>Other names:</b>	«alpha», «alpha»-Diphenylhydrazine N,N-Diphenylhydrazine 1,1-Diphenylhydrazine
<b>Inchi:</b>	InChI=1S/C12H12N2/c13-14(11-7-3-1-4-8-11)12-9-5-2-6-10-12/h1-10H,13H2
<b>InchiKey:</b>	YHYKLKNNBYLTQY-UHFFFAOYSA-N
<b>Formula:</b>	C12H12N2
<b>SMILES:</b>	NN(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	184.24
<b>CAS:</b>	530-50-7

## Physical Properties

Property code	Value	Unit	Source
gf	452.21	kJ/mol	Joback Method
hf	283.37	kJ/mol	Joback Method
hfus	23.14	kJ/mol	Joback Method
hvap	59.54	kJ/mol	Joback Method
log10ws	-3.18		Crippen Method
logp	2.698		Crippen Method
mcvol	152.380	ml/mol	McGowan Method
pc	3659.77	kPa	Joback Method
rinpol	1623.00		NIST Webbook
rinpol	1623.00		NIST Webbook
tb	612.29	K	Joback Method
tc	865.50	K	Joback Method
tf	393.57	K	Joback Method
vc	0.538	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	372.79	J/mol×K	612.29	Joback Method
cpg	388.67	J/mol×K	654.49	Joback Method
cpg	403.16	J/mol×K	696.69	Joback Method
cpg	416.34	J/mol×K	738.89	Joback Method

cpg	428.33	J/mol×K	781.09	Joback Method
cpg	439.23	J/mol×K	823.30	Joback Method
cpg	449.15	J/mol×K	865.50	Joback Method
hvapt	68.80	kJ/mol	497.50	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C530507&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C530507&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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