

# 4-Penten-1-ol, bromoacetate

<b>Inchi:</b>	InChI=1S/C7H11BrO2/c1-2-3-4-5-10-7(9)6-8/h2H,1,3-6H2
<b>InchiKey:</b>	USDNMNTWYGUAAS-UHFFFAOYSA-N
<b>Formula:</b>	C7H11BrO2
<b>SMILES:</b>	C=CCCCOC(=O)CBr
<b>Mol. weight [g/mol]:</b>	207.06

## Physical Properties

Property code	Value	Unit	Source
gf	-123.70	kJ/mol	Joback Method
hf	-280.85	kJ/mol	Joback Method
hfus	20.68	kJ/mol	Joback Method
hvap	46.10	kJ/mol	Joback Method
log10ws	-1.90		Crippen Method
logp	1.891		Crippen Method
mcvol	130.130	ml/mol	McGowan Method
pc	3329.68	kPa	Joback Method
ripol	1146.00		NIST Webbook
ripol	1700.00		NIST Webbook
ripol	1700.00		NIST Webbook
tb	498.69	K	Joback Method
tc	694.72	K	Joback Method
tf	298.85	K	Joback Method
vc	0.494	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	259.14	J/molxK	498.69	Joback Method
cpg	305.41	J/molxK	662.04	Joback Method
cpg	297.07	J/molxK	629.37	Joback Method
cpg	288.29	J/molxK	596.70	Joback Method
cpg	279.05	J/molxK	564.03	Joback Method
cpg	269.33	J/molxK	531.36	Joback Method
cpg	313.32	J/molxK	694.72	Joback Method

dvisc	0.0003045	Paxs	498.69	Joback Method
dvisc	0.0003808	Paxs	465.38	Joback Method
dvisc	0.0004928	Paxs	432.08	Joback Method
dvisc	0.0006659	Paxs	398.77	Joback Method
dvisc	0.0009507	Paxs	365.46	Joback Method
dvisc	0.0014575	Paxs	332.16	Joback Method
dvisc	0.0024579	Paxs	298.85	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R26461&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R26461&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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