

# Octadecanenitrile

<b>Other names:</b>	1-Cyanoheptadecane Heptadecyl cyanide Nitril kyseliny stearove Octadecanonitrile Oktadekannitril Stearonitrile
<b>Inchi:</b>	InChI=1S/C18H35N/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19/h2-17H2,1H3
<b>InchiKey:</b>	RHSBIGNQEIPSCT-UHFFFAOYSA-N
<b>Formula:</b>	C18H35N
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCC#N
<b>Mol. weight [g/mol]:</b>	265.48
<b>CAS:</b>	638-65-3

## Physical Properties

Property code	Value	Unit	Source
gf	233.86	kJ/mol	Joback Method
hf	-249.97	kJ/mol	Joback Method
hfus	43.88	kJ/mol	Joback Method
hvap	66.14	kJ/mol	Joback Method
log10ws	-7.22		Crippen Method
logp	6.771		Crippen Method
mcvol	265.860	ml/mol	McGowan Method
pc	1132.91	kPa	Joback Method
rinpol	348.04		NIST Webbook
rinpol	348.04		NIST Webbook
tb	482.00 ± 3.00	K	NIST Webbook
tb	630.00 ± 4.00	K	NIST Webbook
tc	888.03	K	Joback Method
tf	314.65 ± 1.00	K	NIST Webbook
tf	314.39 ± 0.20	K	NIST Webbook
tf	313.85 ± 1.00	K	NIST Webbook
tf	314.03 ± 0.50	K	NIST Webbook
tf	314.03 ± 0.50	K	NIST Webbook
tf	315.65 ± 1.50	K	NIST Webbook
tf	315.65 ± 0.70	K	NIST Webbook
tf	312.15 ± 2.00	K	NIST Webbook
tf	315.90 ± 0.60	K	NIST Webbook

tf	314.15 ± 2.00	K	NIST Webbook
tf	314.15 ± 2.00	K	NIST Webbook
vc	1.069	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	780.55	J/mol×K	713.32	Joback Method
cpg	798.48	J/mol×K	742.44	Joback Method
cpg	815.58	J/mol×K	771.56	Joback Method
cpg	831.89	J/mol×K	800.68	Joback Method
cpg	847.43	J/mol×K	829.80	Joback Method
cpg	862.23	J/mol×K	858.92	Joback Method
cpg	876.32	J/mol×K	888.03	Joback Method
hfust	56.50	kJ/mol	315.50	NIST Webbook
hvapt	78.60	kJ/mol	554.50	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	547.20	K	13.30	NIST Webbook
tbrp	466.20	K	1.30	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.41565e+01
Coeff. B	-5.04688e+03
Coeff. C	-1.17722e+02
Temperature range (K), min.	481.62
Temperature range (K), max.	688.31

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C638653&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C638653&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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