

# Carbonic acid, neopentyl 3-methylphenyl ester

**Inchi:** InChI=1S/C13H18O3/c1-10-6-5-7-11(8-10)16-12(14)15-9-13(2,3)4/h5-8H,9H2,1-4H3  
**InchiKey:** OMEIGROPJSTJFR-UHFFFAOYSA-N  
**Formula:** C13H18O3  
**SMILES:** Cc1cccc(OC(=O)OCC(C)(C)C)c1  
**Mol. weight [g/mol]:** 222.28

## Physical Properties

Property code	Value	Unit	Source
gf	-174.72	kJ/mol	Joback Method
hf	-472.36	kJ/mol	Joback Method
hfus	19.64	kJ/mol	Joback Method
hvap	57.74	kJ/mol	Joback Method
log10ws	-3.76		Crippen Method
logp	3.557		Crippen Method
mcvol	183.580	ml/mol	McGowan Method
pc	2263.26	kPa	Joback Method
rinpola	1531.00		NIST Webbook
rinpola	1531.00		NIST Webbook
tb	623.98	K	Joback Method
tc	837.28	K	Joback Method
tf	372.02	K	Joback Method
vc	0.686	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	472.53	J/molxK	623.98	Joback Method
cpg	488.51	J/molxK	659.53	Joback Method
cpg	503.52	J/molxK	695.08	Joback Method
cpg	517.57	J/molxK	730.63	Joback Method
cpg	530.70	J/molxK	766.18	Joback Method
cpg	542.93	J/molxK	801.73	Joback Method
cpg	554.30	J/molxK	837.28	Joback Method
dvisc	0.0013196	Paxs	372.02	Joback Method

dvisc	0.0007150	Paxs	414.01	Joback Method
dvisc	0.0004337	Paxs	456.01	Joback Method
dvisc	0.0002862	Paxs	498.00	Joback Method
dvisc	0.0002015	Paxs	539.99	Joback Method
dvisc	0.0001492	Paxs	581.99	Joback Method
dvisc	0.0001151	Paxs	623.98	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357859&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357859&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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