

**sodium chloride**

<b>Other names:</b>	Salt
<b>Inchi:</b>	InChI=1S/ClH.Na/h1H;/q;+1/p-1
<b>InchiKey:</b>	FAPWRFPIFSIZLT-UHFFFAOYSA-M
<b>Formula:</b>	ClNa
<b>SMILES:</b>	[Cl-].[Na+]
<b>Mol. weight [g/mol]:</b>	58.44
<b>CAS:</b>	7647-14-5

## Physical Properties

Property code	Value	Unit	Source
ea	0.73 ± 0.01	eV	NIST Webbook
ea	0.77	eV	NIST Webbook
ea	1.28	eV	NIST Webbook
ie	9.20	eV	NIST Webbook
ie	10.00	eV	NIST Webbook
ie	8.90 ± 0.10	eV	NIST Webbook
ie	8.92 ± 0.06	eV	NIST Webbook
ie	9.80 ± 0.04	eV	NIST Webbook
ie	9.00	eV	NIST Webbook
tf	1074.00	K	Ultrasonic velocity for an equimolar mixture of molten AgI and NaCl in the biphasic region
tf	1074.00	K	Densities of a dissolving mixture of molten (AgI + NaCl)
tt	1074.00	K	Phase-boundary potential in the two-liquid-phase (AgI + NaCl) system

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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rhos	1931.20	kg/m3	1013.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point
rhos	1905.80	kg/m3	1023.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point
rhos	1890.40	kg/m3	1033.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point
rhos	1888.30	kg/m3	1043.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point
rhos	1882.00	kg/m3	1053.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point
rhos	1884.90	kg/m3	1063.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point
rhos	1889.80	kg/m3	1073.00	Density of Crystalline Alkali Chlorides and Their Eutectic Mixtures Near the Melting Point

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.63209e+01
Coeff. B	-1.94159e+04
Coeff. C	-7.90800e+01
Temperature range (K), min.	1073.90
Temperature range (K), max.	1738.20

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Experimental Investigation of the Influence of NaCl on the Vapor-Liquid Equilibrium of Hexane or heptane in Aqueous Salt Solutions from (280 to 340) K: Dependence of Apparent Molar Volumes and Activity Coefficients and on the CO<sub>2</sub> Sequence and Resonance in Solid-C Vapor-Liquid-Liquid-Solid Equilibrium of the water + NaCl + 5-(n-butyl)-5-oxo-1,3-dioxane-2-methyl) Iminodiacetic Acid in Aqueous Sodium Unaturated Solutions from (280 to 353) K: Solubilities and Solution Thermodynamics of a Polyvinyl-LiNa, K, and Na Chloride Complexes in Saline and Aqueous Media: 1,6-dioxane 38.15 K: Hydrophobicity in Aqueous Media for Electrolyte Solutions: Measurement of the Partition Coefficient of Surfactant and STPP in Aqueous Solutions and on Selective Equilibria for the ternary system water + 1,4-dioxane + Formic Acid at 303.15 K: Effects of Na, Ca, Mg, and Al Chloride Salts on Dissolution and Phase Determination of CO<sub>2</sub> Solubility in Water and NaCl Solutions from 280 K to 350 K: Aqueous Solution Conditions for the Formation of Magnesium Chloride with Solid Magnesium Oxide and Carbon Dioxide in Aqueous Sodium Chloride Solutions and on Physicochemical Properties for the Ternary System (NaCl + NaOH + CO<sub>2</sub>) at 298.15 K in the NaCl-CO<sub>2</sub>-H<sub>2</sub>O Ternary System at 298.15 K: Thermodynamic properties of some first and second group metals at various pressures: inorganic salts and organic solvents: Thermodynamics model and parameters:

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## Legend

ea:	Electron affinity
ie:	Ionization energy
vpap:	Vapor pressure
rhos:	Solid Density
tf:	Normal melting (fusion) point
tt:	Triple Point Temperature

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