

# 1-Heptene, 2-methyl-

<b>Other names:</b>	2-Methyl-1-heptene 2-Methylhept-1-ene
<b>Inchi:</b>	InChI=1S/C8H16/c1-4-5-6-7-8(2)3/h2,4-7H2,1,3H3
<b>InchiKey:</b>	RCBGGJURENJHKV-UHFFFAOYSA-N
<b>Formula:</b>	C8H16
<b>SMILES:</b>	C=C(C)CCCC
<b>Mol. weight [g/mol]:</b>	112.21
<b>CAS:</b>	15870-10-7

## Physical Properties

Property code	Value	Unit	Source
gf	95.77	kJ/mol	Joback Method
hf	-92.81	kJ/mol	Joback Method
hfus	13.89	kJ/mol	Joback Method
hvap	39.30	kJ/mol	NIST Webbook
log10ws	-3.02		Crippen Method
logp	3.143		Crippen Method
mvol	119.280	ml/mol	McGowan Method
pc	2659.77	kPa	Joback Method
rinpol	784.00		NIST Webbook
rinpol	783.00		NIST Webbook
rinpol	787.50		NIST Webbook
rinpol	773.00		NIST Webbook
rinpol	785.00		NIST Webbook
rinpol	778.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	784.30		NIST Webbook
rinpol	774.90		NIST Webbook
rinpol	782.00		NIST Webbook
rinpol	783.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	785.00		NIST Webbook
rinpol	776.40		NIST Webbook
rinpol	782.00		NIST Webbook
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rinpol	787.00		NIST Webbook
rinpol	788.00		NIST Webbook
rinpol	785.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	782.00		NIST Webbook
tb	392.50 ± 1.00	K	NIST Webbook
tb	393.15 ± 2.00	K	NIST Webbook
tb	392.65 ± 0.50	K	NIST Webbook
tb	393.15 ± 2.00	K	NIST Webbook
tb	392.45 ± 0.50	K	NIST Webbook
tb	392.75 ± 0.40	K	NIST Webbook
tb	391.65 ± 1.00	K	NIST Webbook
tb	389.65 ± 3.00	K	NIST Webbook
tb	392.00 ± 3.00	K	NIST Webbook
tb	392.50	K	NIST Webbook
tb	392.37 ± 0.30	K	NIST Webbook
tb	392.45 ± 0.50	K	NIST Webbook
tb	391.65 ± 1.00	K	NIST Webbook
tc	567.50	K	Gas-Liquid Critical Temperatures of Some Alkenes, Amines, and Cyclic Hydrocarbons
tf	183.01 ± 0.30	K	NIST Webbook
tf	183.05 ± 0.50	K	NIST Webbook
tf	183.01 ± 0.30	K	NIST Webbook
tf	183.01 ± 1.00	K	NIST Webbook
vc	0.466	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	217.72	J/mol×K	379.00	Joback Method
cpg	230.42	J/mol×K	407.34	Joback Method
cpg	242.63	J/mol×K	435.68	Joback Method

cpg	254.35	J/mol×K	464.02	Joback Method
cpg	265.60	J/mol×K	492.37	Joback Method
cpg	276.40	J/mol×K	520.71	Joback Method
cpg	286.75	J/mol×K	549.05	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.41848e+01
Coeff. B	-3.23383e+03
Coeff. C	-5.39600e+01
Temperature range (K), min.	286.66
Temperature range (K), max.	418.41

## Sources

The Yaws Handbook of Vapor

Pressure:

Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Gas-Liquid Critical Temperatures of  
Some Alkenes, Amines, and Cyclic

Hydrocarbons:

<https://www.doi.org/10.1021/je0341357>

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

KDB:

<https://www.cheric.org/files/research/kdb/mol/mol256.mol>

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C15870107&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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