

# methyl trans-1-propenyl trisulfide

<b>Other names:</b>	Methyl 1-propenyl trisulfide, (E)- Methyl 1-propenyl trisulphide, trans Trisulfide, methyl-1-propenyl, (E)- Trisulfide, methyl-(E)-1-propenyl
<b>Inchi:</b>	InChI=1S/C4H8S3/c1-3-4-6-7-5-2/h3-4H,1-2H3/b4-3+
<b>InchiKey:</b>	WPRUFZZPIFLBDG-ONEGZZNKSA-N
<b>Formula:</b>	C4H8S3
<b>SMILES:</b>	CC=CSSSC
<b>Mol. weight [g/mol]:</b>	152.30

## Physical Properties

Property code	Value	Unit	Source
gf	162.38	kJ/mol	Joback Method
hf	116.94	kJ/mol	Joback Method
hfus	18.71	kJ/mol	Joback Method
hvap	44.91	kJ/mol	Joback Method
log10ws	-3.48		Crippen Method
logp	3.179		Crippen Method
mcvol	111.970	ml/mol	McGowan Method
pc	4450.38	kPa	Joback Method
rinpol	1144.00		NIST Webbook
rinpol	1143.00		NIST Webbook
rinpol	1158.00		NIST Webbook
rinpol	1158.00		NIST Webbook
rinpol	1130.60		NIST Webbook
rinpol	1140.00		NIST Webbook
rinpol	1145.00		NIST Webbook
rinpol	1146.00		NIST Webbook
rinpol	1146.00		NIST Webbook
tb	501.42	K	Joback Method
tc	756.84	K	Joback Method
tf	232.96	K	Joback Method
vc	0.402	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	198.58	J/mol×K	501.42	Joback Method
cpg	208.10	J/mol×K	543.99	Joback Method
cpg	217.06	J/mol×K	586.56	Joback Method
cpg	225.46	J/mol×K	629.13	Joback Method
cpg	233.29	J/mol×K	671.70	Joback Method
cpg	240.57	J/mol×K	714.27	Joback Method
cpg	247.28	J/mol×K	756.84	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R53383&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R53383&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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