

# Hexadecanoic acid, butyl ester

<b>Other names:</b>	Butyl ester of hexadecanoic acid Butyl hexadecanoate Butyl palmitate Palmitic acid, butyl ester n-Butyl hexadecanoate n-Butyl palmitate
<b>Inchi:</b>	InChI=1S/C20H40O2/c1-3-5-7-8-9-10-11-12-13-14-15-16-17-18-20(21)22-19-6-4-2/h3-19
<b>InchiKey:</b>	GLYJVQDYLFUFC-UHFFFAOYSA-N
<b>Formula:</b>	C20H40O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCC(=O)OCCCC
<b>Mol. weight [g/mol]:</b>	312.53
<b>CAS:</b>	111-06-8

## Physical Properties

Property code	Value	Unit	Source
gf	-116.40	kJ/mol	Joback Method
hf	-700.93	kJ/mol	Joback Method
hfus	50.34	kJ/mol	Joback Method
hvap	69.27	kJ/mol	Joback Method
log10ws	-7.06		Crippen Method
logp	6.811		Crippen Method
mcvol	300.100	ml/mol	McGowan Method
pc	1045.97	kPa	Joback Method
rinpol	2157.00		NIST Webbook
rinpol	2188.00		NIST Webbook
rinpol	2157.00		NIST Webbook
rinpol	2188.00		NIST Webbook
rinpol	2174.00		NIST Webbook
rinpol	2174.00		NIST Webbook
rinpol	2169.00		NIST Webbook
rinpol	2157.00		NIST Webbook
ripol	2419.00		NIST Webbook
tb	733.29	K	Joback Method
tc	904.52	K	Joback Method
tf	387.32	K	Joback Method
vc	1.179	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1013.17	J/molxK	904.52	Joback Method
cpg	997.86	J/molxK	875.99	Joback Method
cpg	981.74	J/molxK	847.45	Joback Method
cpg	964.76	J/molxK	818.91	Joback Method
cpg	946.91	J/molxK	790.37	Joback Method
cpg	928.17	J/molxK	761.83	Joback Method
cpg	908.52	J/molxK	733.29	Joback Method
dvisc	0.0016619	Paxs	387.32	Joback Method
dvisc	0.0000733	Paxs	733.29	Joback Method
dvisc	0.0000988	Paxs	675.63	Joback Method
dvisc	0.0001407	Paxs	617.97	Joback Method
dvisc	0.0002156	Paxs	560.30	Joback Method
dvisc	0.0003644	Paxs	502.64	Joback Method
dvisc	0.0007053	Paxs	444.98	Joback Method
hvapt	93.80	kJ/mol	368.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.02510e+01
Coeff. B	-7.76715e+03
Coeff. C	-1.28730e+02
Temperature range (K), min.	517.80
Temperature range (K), max.	648.64

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C111068&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C111068&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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