

# N-Acetyl-N-(3-[(2-ethylhexyl)oxy]propyl)acetamide

<b>Other names:</b>	1-Propanamine, 3-[(2-ethylhexyl)oxy]-, N,N-diacetate 2-Ethylhexyl 3-di-acetaminopropyl ether 3-(2-Ethylhexoxy)propan-1-amine, N,N-diacetyl N-Acetyl-N-{3-[(2-ethylhexyl)oxy]propyl}acetamide
<b>Inchi:</b>	InChI=1S/C15H29NO3/c1-5-7-9-15(6-2)12-19-11-8-10-16(13(3)17)14(4)18/h15H,5-12H2
<b>InchiKey:</b>	WZCSCPCGOWIXKI-UHFFFAOYSA-N
<b>Formula:</b>	C15H29NO3
<b>SMILES:</b>	CCCCC(CC)COCCCN(C(C)=O)C(C)=O
<b>Mol. weight [g/mol]:</b>	271.40

## Physical Properties

Property code	Value	Unit	Source
gf	-179.08	kJ/mol	Joback Method
hf	-648.06	kJ/mol	Joback Method
hfus	38.49	kJ/mol	Joback Method
hvap	66.54	kJ/mol	Joback Method
log10ws	-3.07		Crippen Method
logp	3.004		Crippen Method
mcvol	241.200	ml/mol	McGowan Method
pc	1546.35	kPa	Joback Method
rinpol	1890.00		NIST Webbook
rinpol	1890.00		NIST Webbook
tb	684.76	K	Joback Method
tc	861.70	K	Joback Method
tf	398.37	K	Joback Method
vc	0.917	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	688.79	J/molxK	684.76	Joback Method
cpg	705.58	J/molxK	714.25	Joback Method
cpg	721.55	J/molxK	743.74	Joback Method
cpg	736.70	J/molxK	773.23	Joback Method

cpg	751.06	J/mol×K	802.72	Joback Method
cpg	764.66	J/mol×K	832.21	Joback Method
cpg	777.50	J/mol×K	861.70	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373429&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373429&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/27-590-7/N-Acetyl-N-3-2-ethylhexyl-oxy-propyl-acetamide.pdf>

Generated by Cheméo on 2024-05-13 09:52:36.885169158 +0000 UTC m=+17883205.805746470.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.