

2,6-Dimethylphenyl acetate

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|-----------------------------|---|
| Inchi: | InChI=1S/C10H12O2/c1-7-5-4-6-8(2)10(7)12-9(3)11/h4-6H,1-3H3 |
| InchiKey: | UDDBRJGGGUMTQZ-UHFFFAOYSA-N |
| Formula: | C10H12O2 |
| SMILES: | CC(=O)Oc1c(C)cccc1C |
| Mol. weight [g/mol]: | 164.20 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | -107.45 | kJ/mol | Joback Method |
| hf | -280.94 | kJ/mol | Joback Method |
| hfus | 17.71 | kJ/mol | Joback Method |
| hvap | 50.61 | kJ/mol | Joback Method |
| log10ws | -2.74 | | Crippen Method |
| logp | 2.229 | | Crippen Method |
| mcvol | 135.440 | ml/mol | McGowan Method |
| pc | 2982.79 | kPa | Joback Method |
| rinpol | 1211.00 | | NIST Webbook |
| rinpol | 1211.00 | | NIST Webbook |
| tb | 541.13 | K | Joback Method |
| tc | 756.04 | K | Joback Method |
| tf | 326.08 | K | Joback Method |
| vc | 0.511 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 299.97 | J/molxK | 541.13 | Joback Method |
| cpg | 358.27 | J/molxK | 720.22 | Joback Method |
| cpg | 347.89 | J/molxK | 684.40 | Joback Method |
| cpg | 336.88 | J/molxK | 648.58 | Joback Method |
| cpg | 325.22 | J/molxK | 612.77 | Joback Method |
| cpg | 312.92 | J/molxK | 576.95 | Joback Method |
| cpg | 368.03 | J/molxK | 756.04 | Joback Method |
| dvisc | 0.0002040 | Paxs | 541.13 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0002500 | Paxs | 505.29 | Joback Method |
| dvisc | 0.0003161 | Paxs | 469.45 | Joback Method |
| dvisc | 0.0004153 | Paxs | 433.61 | Joback Method |
| dvisc | 0.0005733 | Paxs | 397.76 | Joback Method |
| dvisc | 0.0008436 | Paxs | 361.92 | Joback Method |
| dvisc | 0.0013513 | Paxs | 326.08 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U373478&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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