

# Pentane, 1,1,1,5-tetrachloro-

<b>Other names:</b>	1,1,1,5-Tetrachloropentane
<b>Inchi:</b>	InChI=1S/C5H8Cl4/c6-4-2-1-3-5(7,8)9/h1-4H2
<b>InchiKey:</b>	VZEWJVRACUZHQR-UHFFFAOYSA-N
<b>Formula:</b>	C5H8Cl4
<b>SMILES:</b>	C1CCCC(C1)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	209.93
<b>CAS:</b>	2467-10-9

## Physical Properties

Property code	Value	Unit	Source
gf	-53.66	kJ/mol	Joback Method
hf	-218.24	kJ/mol	Joback Method
hfus	18.08	kJ/mol	Joback Method
hvap	42.97	kJ/mol	Joback Method
log10ws	-3.63		Crippen Method
logp	3.766		Crippen Method
mcvol	130.270	ml/mol	McGowan Method
pc	2995.87	kPa	Joback Method
rinpol	1211.00		NIST Webbook
rinpol	1220.00		NIST Webbook
rinpol	1211.00		NIST Webbook
tb	460.29	K	Joback Method
tc	668.54	K	Joback Method
tf	268.21	K	Joback Method
vc	0.500	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	270.30	J/mol×K	668.54	Joback Method
cpg	225.21	J/mol×K	460.29	Joback Method
cpg	234.22	J/mol×K	495.00	Joback Method
cpg	242.58	J/mol×K	529.71	Joback Method
cpg	250.32	J/mol×K	564.42	Joback Method

cpg	257.50	J/mol×K	599.12	Joback Method
cpg	264.15	J/mol×K	633.83	Joback Method
dvisc	0.0003739	Paxs	460.29	Joback Method
dvisc	0.0053237	Paxs	268.21	Joback Method
dvisc	0.0027007	Paxs	300.22	Joback Method
dvisc	0.0015615	Paxs	332.24	Joback Method
dvisc	0.0009941	Paxs	364.25	Joback Method
dvisc	0.0006807	Paxs	396.26	Joback Method
dvisc	0.0004933	Paxs	428.28	Joback Method
hvapt	61.70	kJ/mol	386.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.49623e+01
Coeff. B	-4.22683e+03
Coeff. C	-7.59800e+01
Temperature range (K), min.	364.01
Temperature range (K), max.	513.96

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2467109&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2467109&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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