

# (Z,E)-1,3,11-tridecatriene-5,7,9-triyne

<b>Other names:</b>	(Z,E)-1,3,11-Tridecatrien-5,7,9-triyne 1,3(Z),11(E)-Tridecatrien-5,7,9-triyne
<b>Inchi:</b>	InChI=1S/C13H10/c1-3-5-7-9-11-13-12-10-8-6-4-2/h3-7H,1H2,2H3/b6-4+,7-5-
<b>InchiKey:</b>	KAGUESUDHDXNCN-GUBXDBFYSA-N
<b>Formula:</b>	C13H10
<b>SMILES:</b>	C=CC=CC#CC#CC#CC=CC
<b>Mol. weight [g/mol]:</b>	166.22

## Physical Properties

Property code	Value	Unit	Source
gf	915.26	kJ/mol	Joback Method
hf	865.12	kJ/mol	Joback Method
hfus	37.92	kJ/mol	Joback Method
hvap	50.23	kJ/mol	Joback Method
log10ws	-4.21		Crippen Method
logp	2.315		Crippen Method
mcvol	155.330	ml/mol	McGowan Method
pc	2953.69	kPa	Joback Method
rinpol	1656.00		NIST Webbook
rinpol	1656.00		NIST Webbook
rinpol	1656.00		NIST Webbook
rinpol	1656.00		NIST Webbook
ripol	2392.00		NIST Webbook
ripol	2392.00		NIST Webbook
ripol	2392.00		NIST Webbook
ripol	2394.00		NIST Webbook
tb	528.84	K	Joback Method
tc	787.02	K	Joback Method
tf	542.65	K	Joback Method
vc	0.591	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	301.84	J/mol×K	528.84	Joback Method
cpg	315.90	J/mol×K	571.87	Joback Method
cpg	328.98	J/mol×K	614.90	Joback Method
cpg	341.16	J/mol×K	657.93	Joback Method
cpg	352.54	J/mol×K	700.96	Joback Method
cpg	363.21	J/mol×K	743.99	Joback Method
cpg	373.25	J/mol×K	787.02	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R54787&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R54787&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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