

# 4-Nitrobenzoic acid, 2-butyl ester

<b>Other names:</b>	Benzoic acid, 4-nitro, 1-methylpropyl ester Benzoic acid, 4-nitro, 1-methylpropyll ester
<b>Inchi:</b>	InChI=1S/C11H13NO4/c1-3-8(2)16-11(13)9-4-6-10(7-5-9)12(14)15/h4-8H,3H2,1-2H3
<b>InchiKey:</b>	OQECQIDEZLGVDR-UHFFFAOYSA-N
<b>Formula:</b>	C11H13NO4
<b>SMILES:</b>	CCC(C)OC(=O)c1ccc([N+](=O)[O-])cc1
<b>Mol. weight [g/mol]:</b>	223.23
<b>CAS:</b>	39994-84-8

## Physical Properties

Property code	Value	Unit	Source
gf	-56.29	kJ/mol	Joback Method
hf	-306.15	kJ/mol	Joback Method
hfus	28.52	kJ/mol	Joback Method
hvap	68.38	kJ/mol	Joback Method
log10ws	-3.73		Crippen Method
logp	2.550		Crippen Method
mcvol	166.950	ml/mol	McGowan Method
pc	2793.56	kPa	Joback Method
ripol	1628.00		NIST Webbook
ripol	1652.00		NIST Webbook
ripol	1634.00		NIST Webbook
ripol	1633.00		NIST Webbook
ripol	2368.00		NIST Webbook
ripol	2382.00		NIST Webbook
ripol	2413.00		NIST Webbook
ripol	2353.00		NIST Webbook
ripol	2353.00		NIST Webbook
tb	710.43	K	Joback Method
tc	947.65	K	Joback Method
tf	453.44	K	Joback Method
vc	0.643	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	446.27	J/molxK	710.43	Joback Method
cpg	459.28	J/molxK	749.97	Joback Method
cpg	471.29	J/molxK	789.50	Joback Method
cpg	482.33	J/molxK	829.04	Joback Method
cpg	492.43	J/molxK	868.58	Joback Method
cpg	501.61	J/molxK	908.12	Joback Method
cpg	509.90	J/molxK	947.65	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C39994848&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C39994848&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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