

# Dimethylmalonic acid, monochloride, 3-methylbutyl ester

|                      |                                                                       |
|----------------------|-----------------------------------------------------------------------|
| Inchi:               | InChI=1S/C10H17ClO3/c1-7(2)5-6-14-9(13)10(3,4)8(11)12/h7H,5-6H2,1-4H3 |
| InchiKey:            | IZBONMRVMDHJDM-UHFFFAOYSA-N                                           |
| Formula:             | C10H17ClO3                                                            |
| SMILES:              | CC(C)CCOC(=O)C(C)(C)C(=O)Cl                                           |
| Mol. weight [g/mol]: | 220.69                                                                |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -341.05 | kJ/mol  | Joback Method  |
| hf            | -636.88 | kJ/mol  | Joback Method  |
| hfus          | 19.30   | kJ/mol  | Joback Method  |
| hvap          | 56.46   | kJ/mol  | Joback Method  |
| log10ws       | -2.32   |         | Crippen Method |
| logp          | 2.367   |         | Crippen Method |
| mcvol         | 173.010 | ml/mol  | McGowan Method |
| pc            | 2313.61 | kPa     | Joback Method  |
| rinpol        | 1246.00 |         | NIST Webbook   |
| rinpol        | 1246.00 |         | NIST Webbook   |
| tb            | 592.12  | K       | Joback Method  |
| tc            | 791.38  | K       | Joback Method  |
| tf            | 341.89  | K       | Joback Method  |
| vc            | 0.657   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 422.07    | J/mol×K | 592.12          | Joback Method |
| cpg           | 435.71    | J/mol×K | 625.33          | Joback Method |
| cpg           | 448.58    | J/mol×K | 658.54          | Joback Method |
| cpg           | 460.71    | J/mol×K | 691.75          | Joback Method |
| cpg           | 472.11    | J/mol×K | 724.96          | Joback Method |
| cpg           | 482.81    | J/mol×K | 758.17          | Joback Method |
| cpg           | 492.83    | J/mol×K | 791.38          | Joback Method |
| dvisc         | 0.0033109 | Pa×s    | 341.89          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0015851 | Pa×s | 383.59 | Joback Method |
| dvisc | 0.0008768 | Pa×s | 425.30 | Joback Method |
| dvisc | 0.0005391 | Pa×s | 467.00 | Joback Method |
| dvisc | 0.0003590 | Pa×s | 508.71 | Joback Method |
| dvisc | 0.0002543 | Pa×s | 550.41 | Joback Method |
| dvisc | 0.0001890 | Pa×s | 592.12 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361607&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361607&amp;Units=SI</a> |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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