

# 2,6-Dimethoxybenzoquinone

<b>Other names:</b>	2,6-Dimethoxy-p-benzoquinone 2,6-Dimethoxy-1,4-benzoquinone 2,5-Cyclohexadiene-1,4-dione, 2,6-dimethoxy- p-Benzoquinone, 2,6-dimethoxy- 2,6-Dimethoxy-p-quinone 2,6-Dimethoxyquinone 56336 2,6-Dimethoxybenzo-1,4-quinone 2,6-Dimethoxy-2,5-cyclohexadiene-1,4-dione 3,5-Dimethoxy-1,4-benzoquinone NSC 24500 Quinone, 2,6-dimethoxy-
<b>Inchi:</b>	InChI=1S/C8H8O4/c1-11-6-3-5(9)4-7(12-2)8(6)10/h3-4H,1-2H3
<b>InchiKey:</b>	OLBNOBQOQZRLMP-UHFFFAOYSA-N
<b>Formula:</b>	C8H8O4
<b>SMILES:</b>	<chem>COC1=CC(=O)C=C(OC)C1=O</chem>
<b>Mol. weight [g/mol]:</b>	168.15
<b>CAS:</b>	530-55-2

## Physical Properties

Property code	Value	Unit	Source
ea	1.72 ± 0.06	eV	NIST Webbook
gf	-365.88	kJ/mol	Joback Method
hf	-581.01	kJ/mol	Joback Method
hfus	10.30	kJ/mol	Joback Method
hvap	49.36	kJ/mol	Joback Method
log10ws	-0.50		Crippen Method
logp	0.199		Crippen Method
mcvol	119.000	ml/mol	McGowan Method
pc	3585.64	kPa	Joback Method
rinpol	1506.00		NIST Webbook
rinpol	1530.00		NIST Webbook
rinpol	1515.00		NIST Webbook
rinpol	1522.00		NIST Webbook
tb	595.42	K	Joback Method
tc	832.63	K	Joback Method
tf	399.00	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	282.34	J/mol×K	595.42	Joback Method
cpg	295.17	J/mol×K	634.95	Joback Method
cpg	307.48	J/mol×K	674.49	Joback Method
cpg	319.18	J/mol×K	714.02	Joback Method
cpg	330.18	J/mol×K	753.56	Joback Method
cpg	340.39	J/mol×K	793.09	Joback Method
cpg	349.70	J/mol×K	832.63	Joback Method

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C530552&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307I>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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