

Isophthalic acid, 3,7-dimethyloct-6-enyl propyl ester

Inchi:	InChI=1S/C21H30O4/c1-5-13-24-20(22)18-10-7-11-19(15-18)21(23)25-14-12-17(4)9-6-8
InchiKey:	WBWMGSFDLSWXCI-UHFFFAOYSA-N
Formula:	C21H30O4
SMILES:	CCCOC(=O)c1cccc(C(=O)OCCC(C)CCC=C(C)C)c1
Mol. weight [g/mol]:	346.46

Physical Properties

Property code	Value	Unit	Source
gf	-169.89	kJ/mol	Joback Method
hf	-639.16	kJ/mol	Joback Method
hfus	44.74	kJ/mol	Joback Method
hvap	83.24	kJ/mol	Joback Method
log10ws	-6.17		Crippen Method
logp	5.183		Crippen Method
mcvol	293.570	ml/mol	McGowan Method
pc	1311.80	kPa	Joback Method
rinpol	2492.00		NIST Webbook
tb	867.72	K	Joback Method
tc	1075.69	K	Joback Method
tf	475.65	K	Joback Method
vc	1.127	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	910.41	J/molxK	867.72	Joback Method
cpg	926.54	J/molxK	902.38	Joback Method
cpg	941.51	J/molxK	937.04	Joback Method
cpg	955.37	J/molxK	971.70	Joback Method
cpg	968.17	J/molxK	1006.36	Joback Method
cpg	979.93	J/molxK	1041.03	Joback Method
cpg	990.70	J/molxK	1075.69	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U356736&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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